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<p>(21) International Application Number: PCT/US90/02690 (22) International Filing Date: 14 May 1990 (14.05.90) (30) Priority data: 374,343 30 June 1989 (30.06.89) US PCT/US89/05203 14 November 1989 (14.11.89) WO (34) Countries for which the regional or international application was filed: US et al. (71) Applicants: THE TRUSTEES OF COLUMBIA UNIVERSITY IN THE CITY OF NEW YORK [US/US]; West 116th Street &amp; Broadway, New York, NY 10027 (US). SLOAN-KETTERING INSTITUTE FOR CANCER RESEARCH [US/US]; 1275 York Avenue, New York, NY 10021 (US).</p>		<p>(72) Inventors: MARKS, Paul, A. ; Beach Hill Road, Bridgewater, CT 06752 (US). RIFKIND, Richard, A. ; 30 Sutton Place, New York, NY 10022 (US). BRESLOW, Ronald ; 275 Broad Avenue, Englewood, NJ 07631 (US). JUR-SIC, Branko ; 100 Morningside Drive, Apt. 3F, New York, NY 10027 (US). (74) Agent: WHITE, John, P.; Cooper &amp; Dunham, 30 Rockefeller Plaza, New York, NY 10112 (US). (81) Designated States: AT (European patent), AU, BE (European patent), CH (European patent), DE (European patent)*, DK, DK (European patent), ES (European patent), FI, FR (European patent), GB (European patent), HU, IT (European patent), JP, KR, LU (European patent), NL (European patent), NO, SE (European patent).  Published <i>With international search report. Before the expiration of the time limit for amending the claims and to be republished in the event of the receipt of amendments.</i></p>
<p>(54) Title: NOVEL POTENT INDUCERS OF TERMINAL DIFFERENTIATION AND METHODS OF USE THEREOF</p> <p>(57) Abstract</p> <p>The invention provides compounds, several of which belong to a class having two or more nonpolar components connected by a polar group and having polar groups on the termini of the compound. The invention also concerns a method of selectively inducing terminal differentiation of neoplastic cells and thereby inhibiting proliferation of such cells which comprises contacting the cells under suitable condition with an amount of the compound effective to selectively induce terminal differentiation. Moreover, the invention provides a method of treating a patient having a tumor characterized by proliferation of neoplastic cells which comprises administering to the patient an amount of the compound effective to selectively induce terminal differentiation of such neoplastic cells, thereby inhibiting their proliferation and suppressing oncogenicity. Lastly, the present invention provides a pharmaceutical composition comprising a pharmaceutically acceptable carrier and the compound in an amount effective less than an amount which would cause toxicity in the patient.</p>		

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**NOVEL POTENT INDUCERS OF  
TERMINAL DIFFERENTIATION AND METHODS OF USE THEREOF**

5 This is a continuation-in-part application of U.S. Serial  
No. 374,343, filed June 30, 1989, which in turn is a  
continuation-in-part of application U.S. Serial No. 270,963,  
10 filed November 14, 1988, the contents of which are hereby  
incorporated by reference into the present application. \

This application also claims the benefit of PCT  
International Application No. PCT/US89/05203, filed November  
14, 1989, under 35 U.S.C. § 119.

15 **Background of the Invention**

Throughout this application various publications are  
referenced by arabic numerals within parentheses. Full  
20 citations for these publications may be found at the end of  
the specification immediately preceding the claims. The  
disclosures of these publications in their entireties are  
hereby incorporated by reference into this application in  
order to more fully describe the state of the art to which  
25 this invention pertains.

Cancer is a disorder in which a population of cells has  
become, in varying degrees, unresponsive to the control  
mechanisms which normally govern proliferation and  
30 differentiation. For many years there have been two  
principal strategies for chemotherapeutic treatment of  
cancer: a) blocking hormone-dependent tumor cell  
proliferation by interference with the production or  
peripheral action of sex hormones; and b) killing cancer  
35 cells directly by exposing them to cytotoxic substances,

which injure both neoplastic and normal cell populations.

5 Relatively recently, cancer therapy is also being attempted  
by the induction of terminal differentiation of the  
neoplastic cells (1). In cell culture models  
differentiation has been reported by exposure of cells to a  
variety of stimuli, including: cyclic AMP and retinoic acid  
10 (2,3), aclarubicin and other anthracyclines (4).

There is abundant evidence that neoplastic transformation  
does not necessarily destroy the potential of cancer cells  
to differentiate (1,5,6). There are many examples of tumor  
15 cells which do not respond to the normal regulators of  
proliferation and appear to be blocked in the expression of  
their differentiation program, and yet can be induced to  
differentiate and cease replicating. A variety of agents,  
including some relatively simple polar compounds (5,7-9),  
20 derivatives of vitamin D and retinoic acid (10-12), steroid  
hormones (13), growth factors (6,14), proteases (15,16),  
tumor promoters (17,18), and inhibitors of DNA or RNA  
synthesis (4,19-24), can induce various transformed cell  
lines and primary human tumor explants to express more  
25 differentiated characteristics.

Early studies by the present inventors identified a series  
of polar compounds that were effective inducers of  
differentiation in a number of transformed cell lines (8,9).  
30 Of these, the most effective inducer, until recently, was  
the hybrid polar/apolar compound N,N'-hexamethylene  
bisacetamide (HMBA) (9). The use of polar/apolar compounds  
to induce murine erythroleukemia cells (MELC) to undergo  
erythroid differentiation with suppression of oncogenicity  
35 has proved a useful model to study inducer-mediated

differentiation of transformed cells (5,7-9). HMBA-induced MELC terminal erythroid differentiation is a multistep process. Upon addition of HMBA to MELC (745A-DS19) in culture, there is a latent period of 10 to 12 hours before commitment to terminal differentiation is detected. Commitment is defined as the capacity of cells to express terminal differentiation despite removal of inducer (25). Upon continued exposure to HMBA there is progressive recruitment of cells to differentiate. Recently, the present inventors reported that MELC cell lines made resistant to relatively low levels of vincristine become markedly more sensitive to the inducing action of HMBA and can be induced to differentiate with little or no latent period (26).

HMBA is capable of inducing phenotypic changes consistent with differentiation in a broad variety of cells lines (5). The characteristics of the drug induced effect have been most extensively studied in the murine erythroleukemia cell system (MELC) (5,25,27,28). MELC induction of differentiation is both time and concentration dependent. The minimum concentration required to demonstrate an effect in vitro in most strains is 2 to 3 mM; the minimum duration of continuous exposure generally required to induce differentiation in a substantial portion (>20%) of the population without continuing drug exposure is about 36 hours.

The primary target of action of HMBA is not known. There is evidence that protein kinase C is involved in the pathway of inducer-mediated differentiation (29). The in vitro studies provided a basis for evaluating the potential of HMBA as a cytodifferentiation agent in the treatment of human cancers

(30). Several phase I clinical trials with HMBA have been completed (31-36). Recently, the first evidence was reported that this compound can induce a therapeutic response in patients with cancer (35,36). These phase I clinical trials demonstrate that the potential efficacy of HMBA is limited, in part, by dose-related toxicity which prevents achieving optimal blood levels and by the need for intravenous administration of large quantities of the agent, over prolonged periods.

The present invention provides new chemical inducers which are 2-20 times more active than HMBA. It has unexpectedly been found that compounds having two or more nonpolar components connected by a polar group and having groups on the termini of the compound are effective inducers of terminal differentiation. For instance, bis-hexamethylene triacetamide, which comprises three acetamide groups connected by two six-carbon chains, has been found to be a potent inducer of terminal differentiation in MELC.

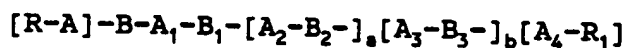
This new class of compounds of the present invention may be useful for selectively inducing terminal differentiation of neoplastic cells and therefore aid in treatment of tumors in patients.

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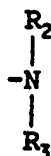
Summary of the Invention

The invention provides a class of compounds having the structure:



wherein each of A, A<sub>1</sub>, A<sub>2</sub>, A<sub>3</sub>, and A<sub>4</sub> represent a polar group which comprises a nitrogen, sulfur or oxygen atom wherein each of A, A<sub>1</sub>, A<sub>2</sub>, A<sub>3</sub>, and A<sub>4</sub> may independently be the same as, or different from, the others of A, A<sub>1</sub>, A<sub>2</sub>, A<sub>3</sub>, and A<sub>4</sub>;

wherein each of R and R<sub>1</sub> is a hydrogen atom; a lower alkyl, alkenyl, or alkynyl group; or a group having the structure:



wherein each of R<sub>2</sub> and R<sub>3</sub> being a hydrogen atom or a lower alkyl, alkenyl, or alkynyl group; and wherein each of R, R<sub>1</sub>, R<sub>2</sub> and R<sub>3</sub> may independently be the same as, or different from the other of R, R<sub>1</sub>, R<sub>2</sub> and R<sub>3</sub>;

wherein each of [R-A] and [A<sub>4</sub>-R<sub>1</sub>] have a dipole moment greater than about 2.7 Debye units;

wherein each of B, B<sub>1</sub>, B<sub>2</sub>, and B<sub>3</sub> represents a nonpolar group which comprises at least 4 atoms in a chain, the termini of which chains are attached to A and A<sub>1</sub>, A<sub>1</sub> and A<sub>2</sub>, A<sub>2</sub> and A<sub>3</sub>, and A<sub>3</sub> and A<sub>4</sub>, respectively; wherein each such atom is oxygen, nitrogen, carbon, or sulfur and wherein each of B, B<sub>1</sub>, B<sub>2</sub>, and B<sub>3</sub> may independently be the same as, or different

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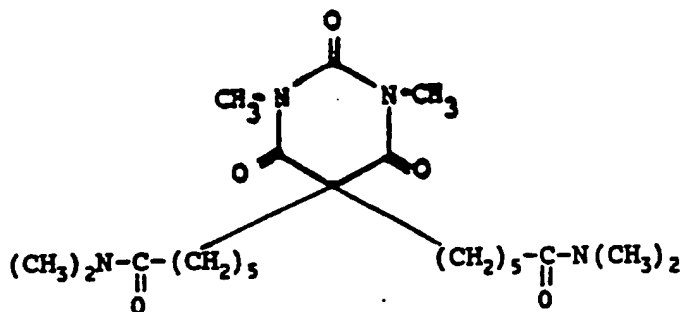
from the others of B, B<sub>1</sub>, B<sub>2</sub> and B<sub>3</sub>;

and wherein each of a and b is independently 0 or 1.

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Other compounds of this invention include a compound having the structure:

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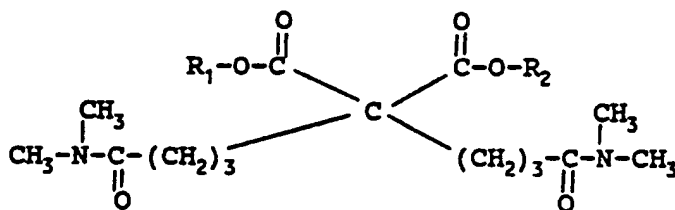


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or a compound having the structure:

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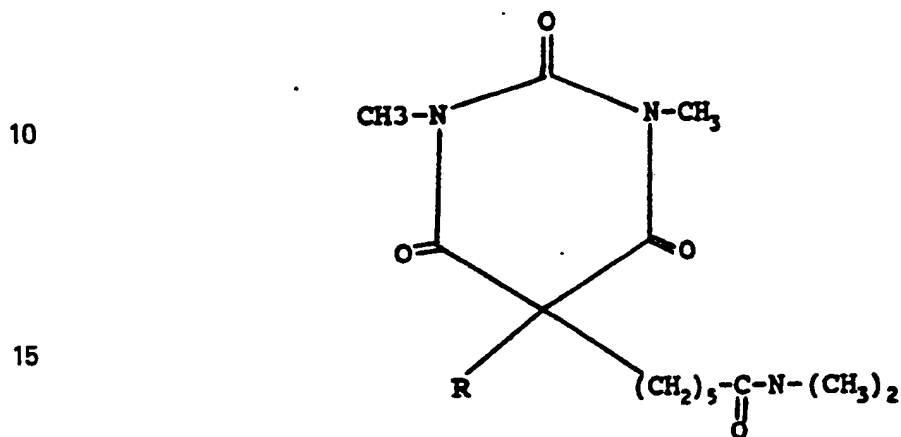
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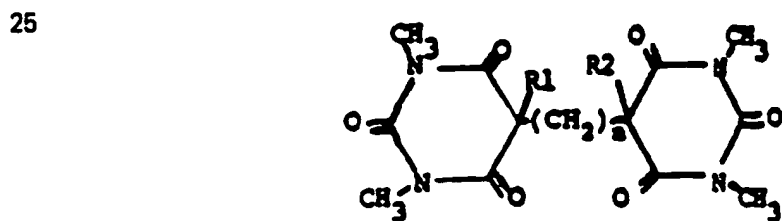
wherein R<sub>1</sub> and R<sub>2</sub> may be the same or different and each is a lower alkyl group.

5 Also provided by this invention is a compound having the structure:



20 wherein R is H or a lower alkyl group.

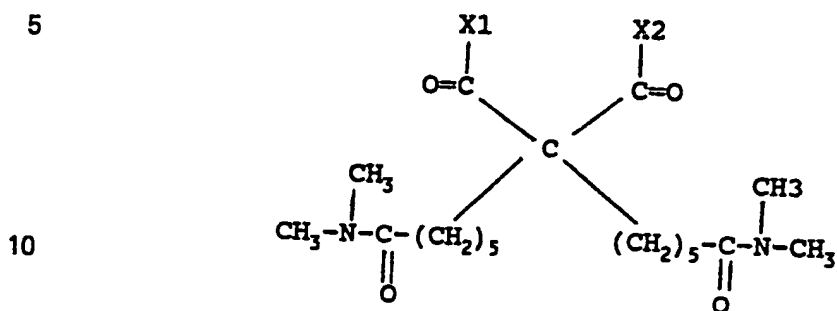
This invention further provides a compound having the structure:



35 wherein n is an integer which is greater than 1 and R<sub>1</sub> and R<sub>2</sub> may be the same or different and each is H or a lower alkyl group.

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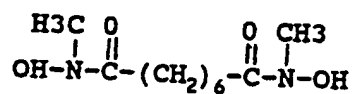
A compound having the structure:



15 is also provided by this invention wherein X1 and X2 may independently be the same or different and each is  $\text{N}(\text{CH}_3)_2$ , NH-phenyl, or  $\text{HNCH}_3$ .

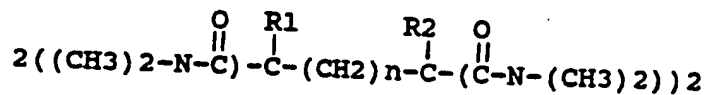
This invention also provides a compound having the structure:

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or a compound having the structure:

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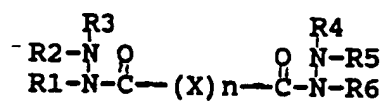
30 wherein R1 and R2 may be the same or different and each is H or a lower alkyl group and n is an integer from 1 to 10.

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This invention further provides a compound having the structure:

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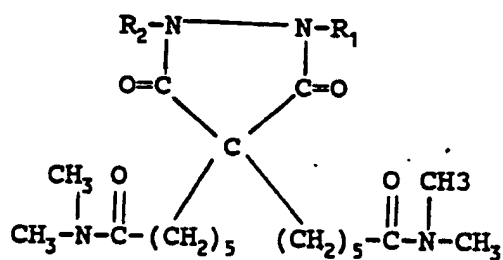
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wherein R1, R2, R3, R4, R5, R6 may independently be the same or different from each other and is H or a lower alkyl group; wherein X is methyl or phenyl; and n is an integer from 1 to about 15.

A compound having the structure:

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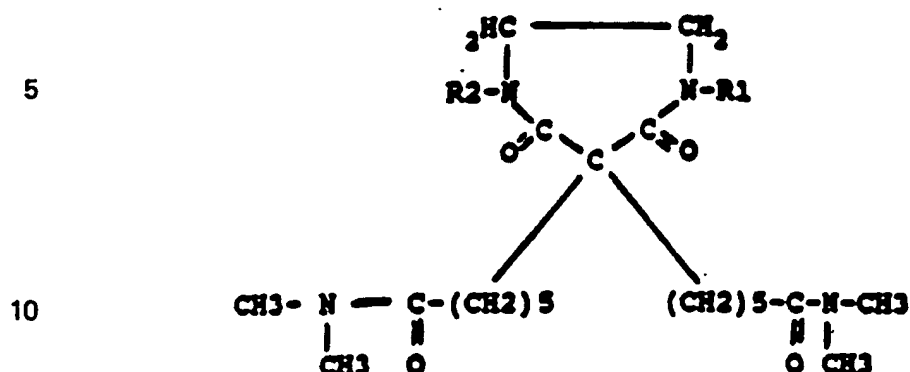
is also provided by this invention wherein R1 and R2 may be the same or different and is H or CH3.

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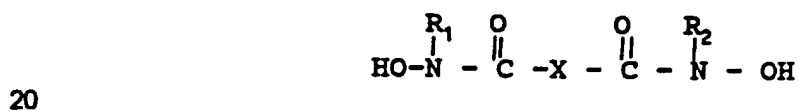
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The invention also provides a compound having the structure:

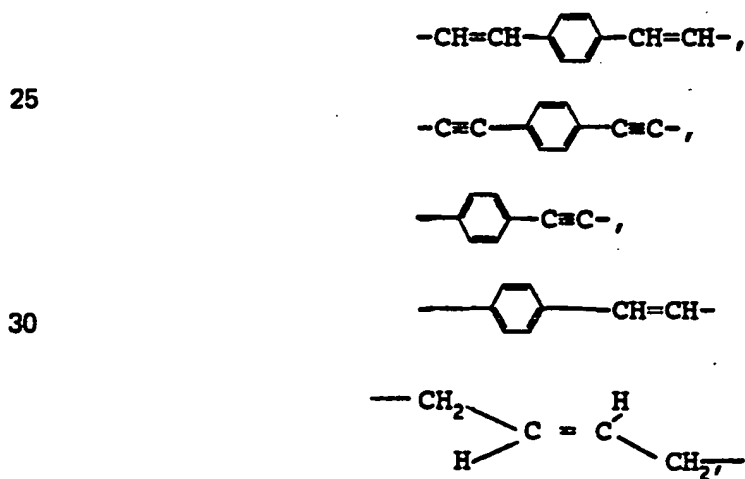


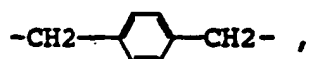
wherein R<sub>1</sub> and R<sub>2</sub> may be the same or different and is H or CH<sub>3</sub>.

A compound having the structure:



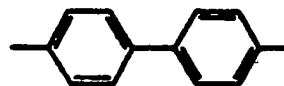
is provided by this invention, wherein X has the structure:





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or



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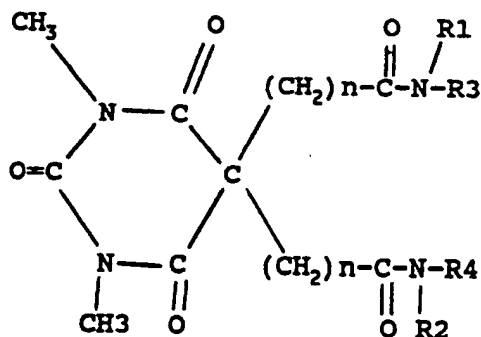


wherein  $R_1$  and  $R_2$  are the same or different and is H or lower alkyl group.

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Further provided by this invention is a compound having the structure:

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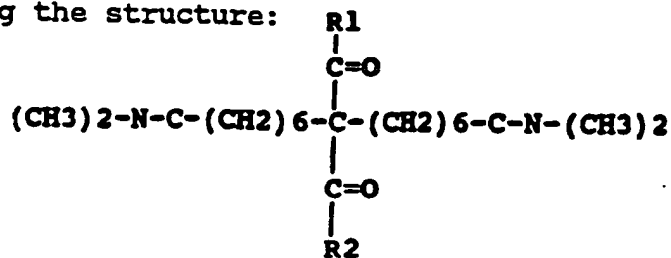


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wherein  $R_1$  and  $R_2$  may independently be the same or different and each may be H or a lower alkyl group; wherein  $R_3$  and  $R_4$  may independently be the same or different and each may be  $\text{CH}_3$  or  $\text{OH}$ ; and wherein  $n$  is 5 or 6.

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A compound having the structure:



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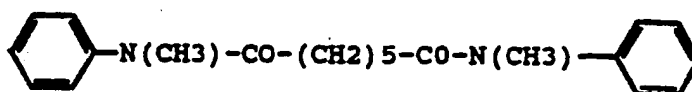
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is also provided by this invention, wherein  $R_1$  and  $R_2$  is the same or different and is hydrogen, lower alkyl, alkenyl, alkynyl, an amide or hydroxyamide.

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This invention further provides a compounds having the following structures:

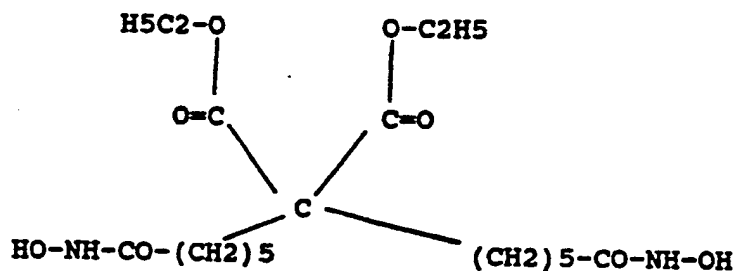
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and

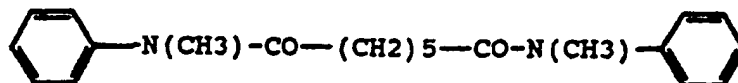
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also,

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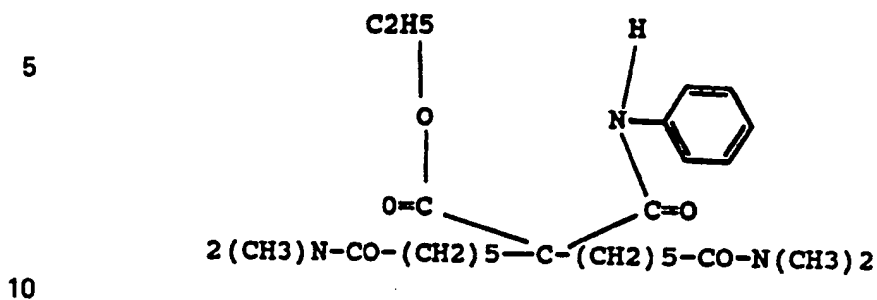


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and a compound having the structure:



15 The invention also concerns a method of selectively inducing terminal differentiation of neoplastic cells and thereby inhibiting proliferation of such cells which comprises contacting the cells under suitable condition with an amount of the compound effective to selectively induce terminal differentiation.

20 Moreover, the invention provides a method of treating a patient having a tumor characterized by proliferation of neoplastic cells which comprises administering to the patient an amount of the compound effective to selectively induce terminal differentiation of such neoplastic cells, thereby inhibiting their proliferation, and suppressing oncogenicity.

25

30 Lastly, the present invention provides a pharmaceutical composition comprising a pharmaceutically acceptable carrier and the compound in an amount less than an amount which cause toxicity in the patient.

Brief Description of the Figures

5     Figure 1: Comparison of the hybrid polar/apolar compounds,  
          (A) hexamethylene bisacetamide (HMBA); (B)  
          suberic acid bis-N-methyl diacetamide (SBDA); and  
          (C) bis-hexamethylene triacetamide (BHTA) as  
10           inducers of differentiation of vincristine-  
          sensitive (745A-DS19) (●) and vincristine-  
          resistant (VCR.C(2) 15) (▲) MELC for HMBA and  
          SBDA and (V3.17) for BHTA. Each compound was  
15           added to the cells in culture at the final  
          concentration indicated. Benzidine reactive  
          cells (left panel of each section of this figure)  
          were determined after 4 d. of culture and  
          commitment (right panel of each section of this  
          figure) after 2 d. of culture.

20     Figure 2: Concentration dependent curves of inducer  
          activity of compound 12 with vincristine-  
          sensitive (745A-DS19) (●) and vincristine-  
          resistant (VCR.C(2)15) (▲) MELC. The compound  
25           was added to the cultures at the final  
          concentrations indicated. Benzidine reactive  
          cells were determined after 5 d. in culture and  
          commitment after 2 d. of culture.

30     Figure 3: Optimal concentration of IC-135 for inducement of  
          terminal differentiation.

Figure 4: Comparison of HMBA and IC-135 on DS19 cells.

35     Figure 5A  
          and B: Comparison of % cell committed for HMBA and IC-  
          135 on V3.17 and DS19 cell lines.



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Figure  
6A and B: Comparison of B+ $\frac{1}{2}$  for HMBA and IC-135 on V3.17  
and DS19 cell lines.

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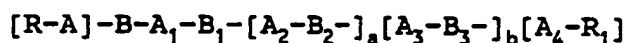
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Detailed Description of the Invention

The invention provides a class of compounds having the  
 5 structure:

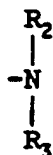


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wherein each A, A<sub>1</sub>, A<sub>2</sub>, A<sub>3</sub>, and A<sub>4</sub> represent a polar group  
 which comprises a nitrogen, sulfur or oxygen atom and  
 wherein each of A, A<sub>1</sub>, A<sub>2</sub>, A<sub>3</sub>, and A<sub>4</sub> may independently be  
 the same as, or different from, the others of A, A<sub>1</sub>, A<sub>2</sub>, A<sub>3</sub>,  
 15 and A<sub>4</sub>;

wherein each of R and R<sub>1</sub> is a hydrogen atom; a lower alkyl,  
 alkenyl, or alkynyl group; or a group having the structure:

20



25 wherein each of R<sub>2</sub> and R<sub>3</sub> being a hydrogen atom or a lower  
 alkyl, alkenyl, or alkynyl group; and wherein each of R, R<sub>1</sub>,  
 R<sub>2</sub> and R<sub>3</sub> may independently be the same as, or different  
 from, the others of R, R<sub>1</sub>, R<sub>2</sub> and R<sub>3</sub>;

30 wherein each of [R-A] and [A<sub>4</sub>-R<sub>1</sub>] have a dipolar moment  
 greater than about 2.7 Debye units;

wherein each of B, B<sub>1</sub>, B<sub>2</sub>, and B<sub>3</sub> represents a nonpolar  
 group which comprises at least 4 atoms in a chain, the  
 35 termini of which chains are attached to A and A<sub>1</sub>, A<sub>1</sub> and A<sub>2</sub>,

$A_2$  and  $A_3$ , and  $A_3$  and  $A_4$ , respectively; wherein each such atom is oxygen, nitrogen, carbon, or sulfur and wherein each of  $B$ ,  $B_1$ ,  $B_2$ , and  $B_3$  may independently be the same as, or  
5 different from the others of  $B$ ,  $B_1$ ,  $B_2$ , and  $B_3$ ;

and wherein each of  $a$  and  $b$  is independently 0 or 1.

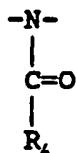
10 The compounds of the present invention are made up of two components. One component comprises a polar group, i.e. functional groups with significant dipole moments, such as amides, sulfoxides, amine oxides and related functional groups.

15 The terminal portions of the compound, represented by  $R-A$  and  $A_4-R_1$ , each have dipole moments greater than about 2.7 debye units. The polar groups within the compound, represented by  $-A_1$ ,  $-A_2-$  and  $-A_3-$ , have significant dipolar moments but not necessarily in excess of 2.7 debye units.  
20 In the preferred embodiments, the polar groups are carbonyl radicals or bivalent radicals of an amide, a sulfoxide or a amine oxide. Each polar group need not necessarily be the same as the other polar groups. In the most preferred  
25 embodiments, the polar groups within the compound are the same as each other and the terminal polar groups are the same. Preferably, all the polar groups are amide groups attached to the compound at the nitrogen atom or at the carbon atom of the carbonyl radical. The amide group may  
30 comprise one or more hydrocarbon substituents, such as a lower alkyl or alkenyl groups, including branched or unbranched groups. The term "lower alkyl or alkenyl group" is intended to include saturated and unsaturated hydrocarbon groups with 1 to about 5 carbon atoms.

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The embodiments where a and b are 0 and A is a carbonyl radical or a group having the structure:

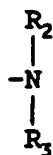
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10 wherein  $\text{R}_4$  is a hydrogen atom or a lower alkyl or alkenyl group, have proven to be most useful embodiments to date.

Particularly preferred are compounds where a and b are 0, A is a carbonyl radical and R has the structure:

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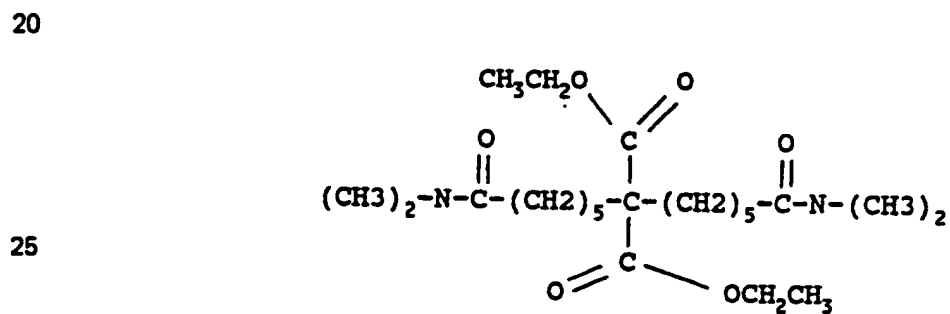
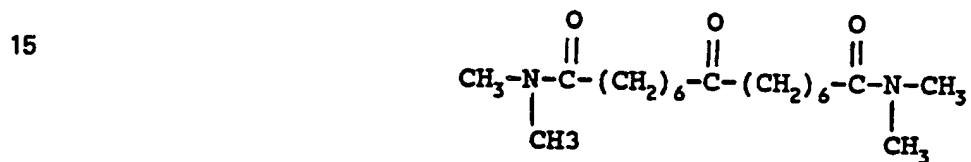
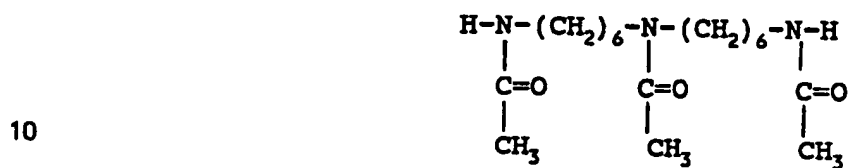
wherein  $\text{R}_2$  and  $\text{R}_3$  each is hydrogen atom, a methyl group or a ethyl group.

25 The compound also requires at least two nonpolar sections, designated B and  $\text{B}_1$ , which are attached to and connect polar groups. Additional nonpolar sections may also be present, e.g.  $\text{B}_2$  when a is 1 and  $\text{B}_3$  when b is 1. The nonpolar sections may comprise linear saturated hydrocarbon chains, 30 linear unsaturated hydrocarbon chains containing one or more double or triple bonds, or saturated or unsaturated hydrocarbon chains containing one or more lower alkyl or alkenyl groups or small carbocyclic rings as substituents. In one of the preferred embodiments, the nonpolar groups are hydrocarbon chains comprising 4 to 7 methylene groups, 35 especially preferred are hydrocarbon chains containing 6

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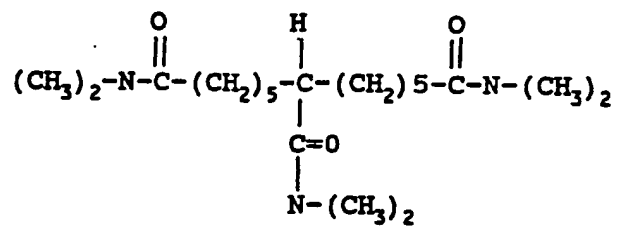
carbon atoms.

Some of the preferred compounds for the practice of the  
 5 present invention are those having the structures:



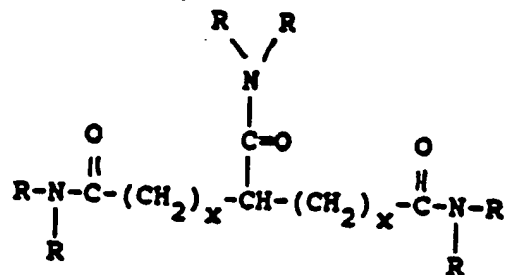
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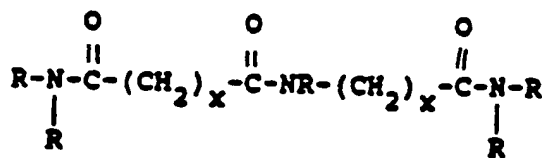
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25

or



30 wherein R is hydrogen or a methyl group and x is 5 or 6.

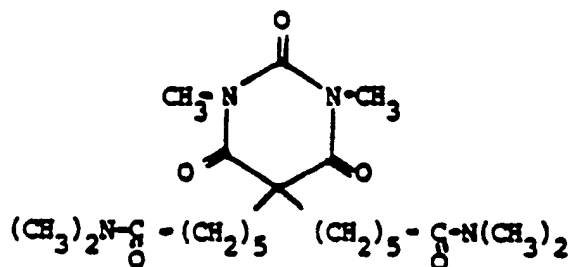
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Other compounds of this invention include a compound having the structure:

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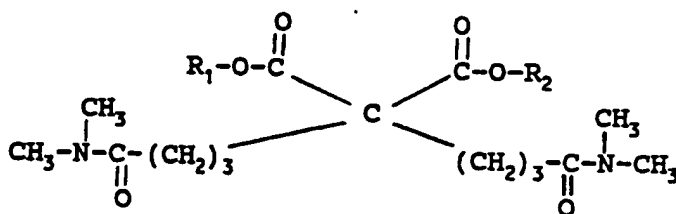


15

or a compound having the structure:

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wherein R1 and R2 may be the same or different and each is a lower alkyl group.

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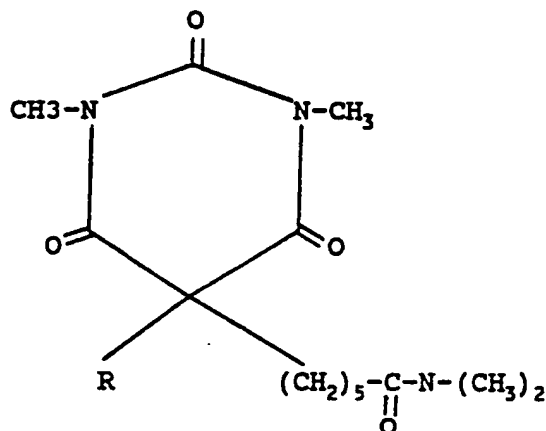
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Also provided by this invention is a compound having the structure:

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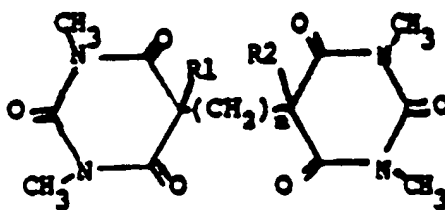


wherein R is H or a lower alkyl group.

This invention further provides a compound having the structure:

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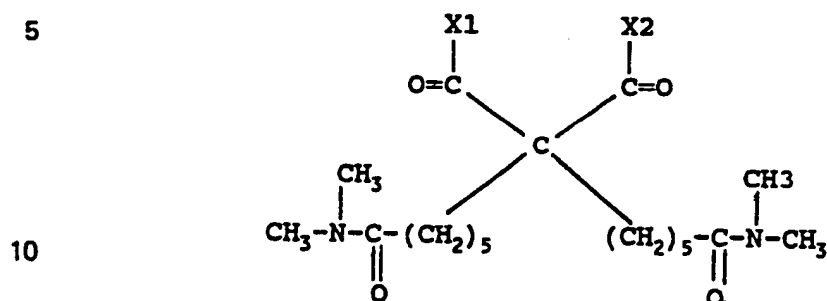
wherein n is an integer which is greater than 1 and R<sub>1</sub> and R<sub>2</sub> may be the same or different and each is H or a lower alkyl group.

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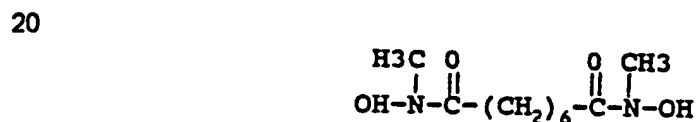
A compound having the structure:



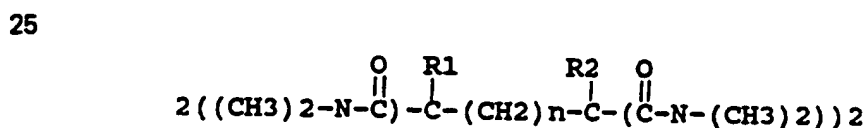
is also provided by this invention wherein X1 and X2 may independently be the same or different and each is  $\text{N}(\text{CH}_3)_2$ , NH-phenyl, or  $\text{HNCH}_3$ .

15

This invention also provides a compound having the structure:



or a compound having the structure:



wherein R1 and R2 may be the same or different and each is H or a lower alkyl group;

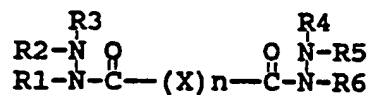
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and n is an integer from 1 to 10.

24

This invention further provides a compound having the structure:

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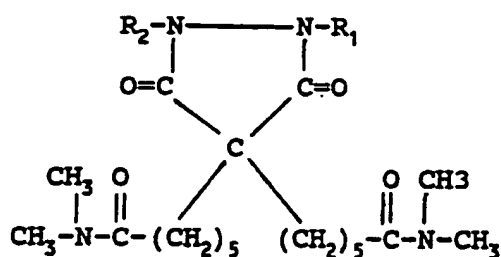
wherein R1, R2, R3, R4, R5, R6 may independently be the same or different from each other and is H or a lower alkyl group;

wherein X is methyl or phenyl; and

n is an integer from 1 to about 15.

A compound having the structure:

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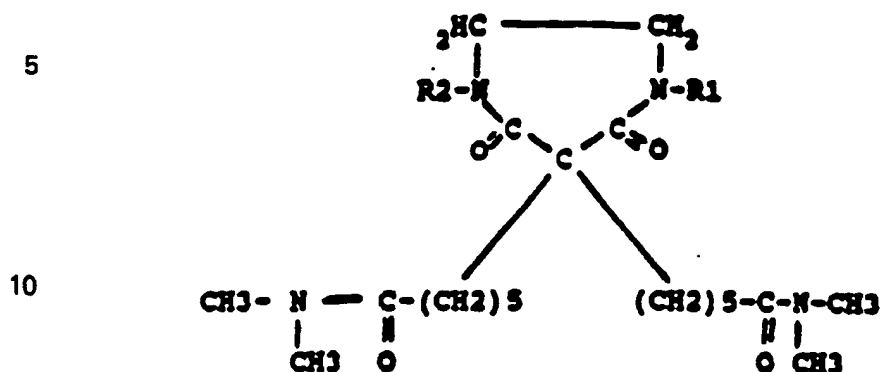
is also provided by this invention wherein R1 and R2 may be the same or different and is H or CH3.

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The invention also provides a compound having the structure:

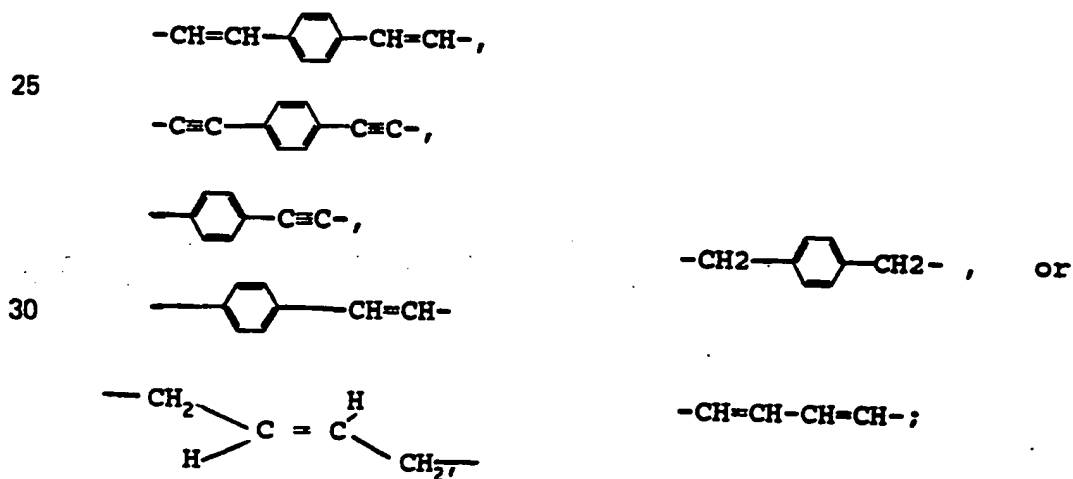


wherein R<sub>1</sub> and R<sub>2</sub> may be the same or different and is H or CH<sub>3</sub>.

A compound having the structure:



is provided by this invention, wherein X has the structure:



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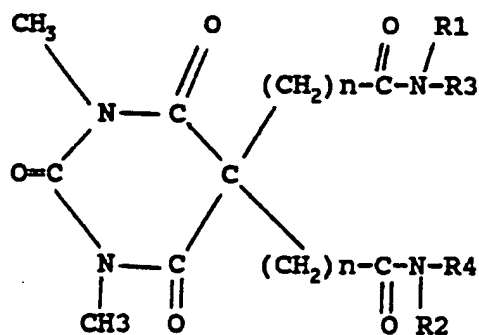
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wherein  $R_1$  and  $R_2$  are the same or different and is H or lower alkyl group.

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Further provided by this invention is a compound having the structure:

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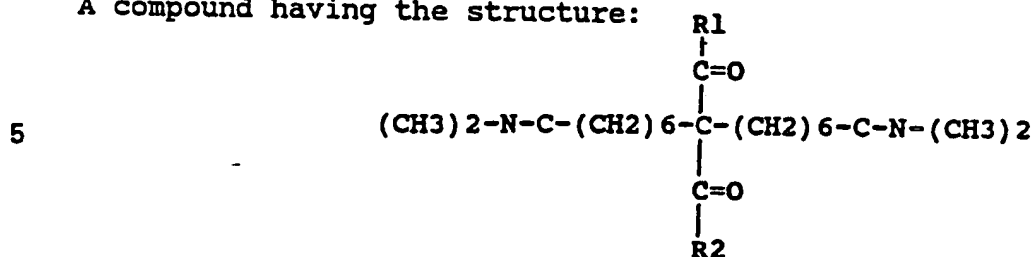
wherein  $R_1$  and  $R_2$  may independently be the same or different and each may be H or a lower alkyl group; wherein  $R_3$  and  $R_4$  may independently be the same or different and each may be  $CH_3$  or  $OH$ ; and wherein  $n$  is 5 or 6.

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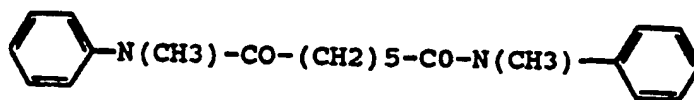
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A compound having the structure:

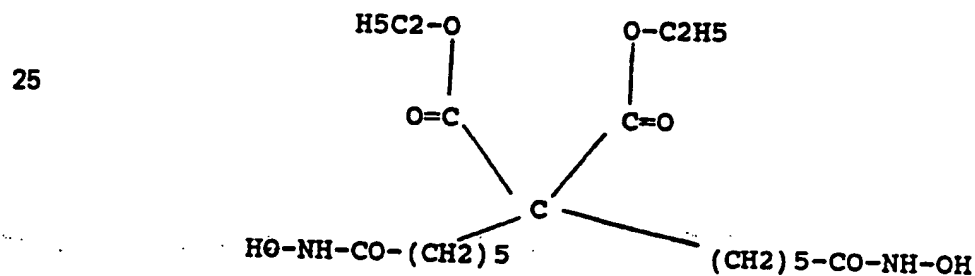


10 is also provided by this invention, wherein  $R_1$  and  $R_2$  is the same or different and is hydrogen, lower alkyl, alkenyl, alkynyl, an amide or hydroxyamide.

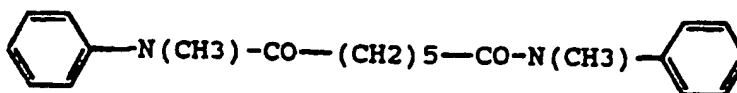
15 This invention further provides a compounds having the following structures:



and

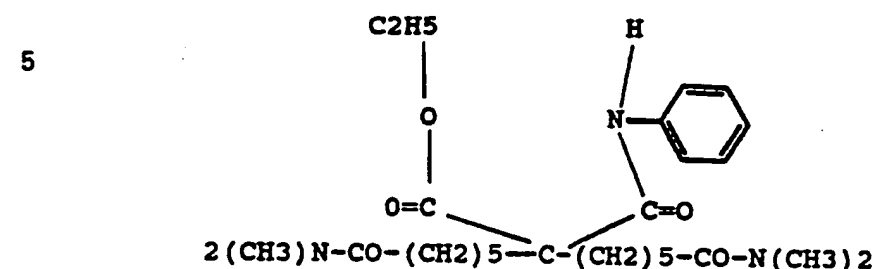


also,



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and a compound having the structure:



15 The invention also concerns a method of selectively inducing terminal differentiation of neoplastic cells and thereby inhibiting proliferation of such cells which comprises contacting the cells under suitable conditions with an amount of the compounds shown above effective to selectively induce terminal differentiation in the cells.

20 A method of selectively inducing terminal differentiation of neoplastic cells and thereby inhibiting proliferation of such cells is further provided by this invention which comprises contacting the cells under suitable conditions with an amount of a compound effective to selectively induce terminal differentiation, the compound having the structure:



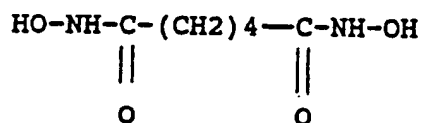
30 wherein X is phenyl or methyl and n is an integer from 1 to 15.

The contacting must be performed continuously for a prolonged period of time, i.e. for at least 48 hours, preferably for about 4-5 days or longer.

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The method may be practiced in vivo or in vitro. If the method is practiced in vitro, contacting may be effected by incubating the cells with the compound. The concentration of the compound in contact with the cells should be from about 1  $\mu$ M to about 25 mM, preferably from about 4  $\mu$ M to about 5 mM.

However, for the compound having the structure:



an amount of about 0.01 mM to about 10 mM has been shown to be effective.

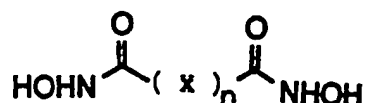
The concentration depends upon the individual compound. For example, compound 12 of Table 1 should have a concentration from about 0.1 to about 2 mM, preferably from about 0.5 to about 0.7 mM. However, for the compounds listed in Table 4, the optimal concentration is as low as 5  $\mu$ M to about 5 mM. Another factor determining the preferable range is the state of the tumor cells. Thus, in cells which have low levels of vincristine resistance, the range of effective concentration of compound 12 from Table 1 is from about 0.01 to about 0.3 mM with a preferable range of about 0.05 to about 0.1 mM.

The invention also concerns a method of treating a patient having a tumor characterized by proliferation of neoplastic cells which comprises administering to the patient an amount

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of the compounds shown above, or with a compound having the structure:

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wherein X is phenyl or methyl and n is an integer from 1 to 15, effective to selectively induce terminal differentiation of such neoplastic cells thereby inhibiting their proliferation, and suppressing oncogenicity.

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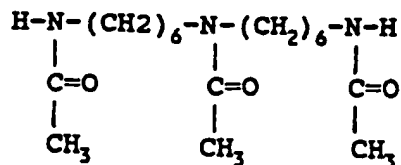
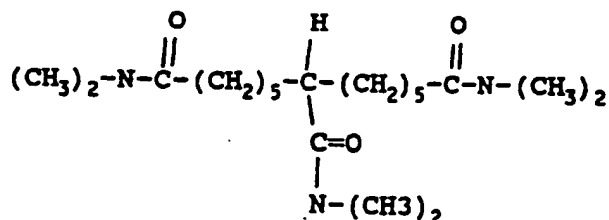
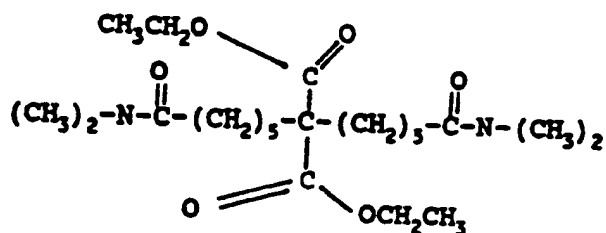
The method of the present invention is intended for the treatment of human patients with tumors. However, it is also likely that the method would be effective in the treatment of tumors in other animals. The term tumor is intended to include any cancer caused by the proliferation of neoplastic cells, such as lung cancer, acute lymphoid myeloma, bladder melanoma, renal carcinoma, breast carcinoma, or colorectal carcinoma. The administration of the compound to the patient may be effected orally or parenterally. To date, administration intravenously has proven to be effective. The administration of the compound must be performed continuously for a prolonged period of time, such as for at least 3 days preferably more than 5 days. In the most preferred embodiments, the administration is effected continuously for at least 10 days and is repeated at intervals wherein at each interval the administration is continuously effected for at least 10 days. For example, the administration may be effected at intervals as short as 5-10 days, up to about 25-35 days and

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continuously for at least 10 days during each such interval. The optimal interval period will vary depending on the type of patient and tumor. For example, in the incidence of acute leukemia, the so called myelodysplastic syndrome, continuous infusion would seem to be indicated so long as the patient tolerated the drug without toxicity and there was a positive response.

10 The amount of the compound administered to the patient is less than an amount which would cause toxicity in the patient. In the certain embodiments, wherein the compounds have the structures:



**or**

the amount of the compound which is administered to the patient is less than the amount which causes a concentration of the compound in the patient's plasma to equal or exceed the toxic level of the compound. Preferably, the concentration of the compound in the patient's plasma is maintained at about 1.0 mM. It has been found with HMBA, that administration of the compound in an amount from about 5 gm/m<sup>2</sup>/day to about 30 gm/m<sup>2</sup>/day, particularly about 20 gm/m<sup>2</sup>/day, is effective without producing toxicity in the patient. For the compound shown above, in vitro studies have shown that the optimal amount of the compounds is substantially lower than 30 gm/m<sup>2</sup>/day, and may even be lower than 1 gm/m<sup>2</sup>/day. The optimal amount of the compound which should be administered to the patient in the practice of the present invention will depend on the particular compound used and the type of cancer being treated.

This invention, in addition to the above listed compounds, is intended to encompass the use of homologs and analogs of such compounds. In this context, homologs are molecules having substantial structural similarities to the above-described compounds and analogs are molecules having substantial biological similarities regardless of structural similarities.

The invention also concerns a pharmaceutical composition comprising a pharmaceutically acceptable carrier, such as sterile pyrogen-free water, and any of the compounds listed above in an amount less than an amount which if administered intravenously or orally to a patient would cause the patient's blood plasma concentration of the compound to exceed toxic levels.

The invention is illustrated in the Experimental Detail section which follows. This section is set forth to aid in an understanding of the invention but is not intended to, and should not be construed to, limit in any way the invention as set forth in the claims which follow thereafter.

### Experimental Details

#### Methods

#### Cells and Materials

MELC 745A-DS19 cells and the variants of MELC derived from this cell line, namely, the vincristine-resistant MELC V3.17 and VCR.C(2)15 cell lines (26), and the dimethylsulfoxide-resistant cell line, DR10 (37), were maintained in alpha minimal essential medium containing 10% fetal calf serum (16). Cell cultures for all experiments were initiated with cells in logarithmic growth phase (day 2 cultured cells) at a density of  $10^5$  cells/ml. Inducer compounds were added in the final concentrations indicated below, dissolved in culture medium without fetal calf serum unless otherwise indicated. Cell density and benzidine reactivity were determined as described (16).

Commitment to terminal differentiation, characterized by limited cell division (colony size <32 cells) and accumulation of hemoglobin (benzidine reactive colonies) was assayed by a colony cloning assay using 2% methylcellulose as described (25).

Table 1

CPD	Structure	Mol Wt.	Optimal Concentration (mM)	Benzidine Reactive Cells (%)	Commitment (%)
1.	$\text{CH}_3-\text{C}(=\text{O})-\text{NH}-(\text{CH}_2)_6-\text{NH}-\text{C}(=\text{O})-\text{CH}_3$	200	5	>95	>95
2.	$\text{CH}_3-\text{C}(=\text{O})-\text{NH}-\text{CH}_3$	75	90	>70	NO
3.	$\text{CH}_3-\text{C}(=\text{O})-\text{NH}-(\text{CH}_2)_2-\text{C}(\text{CH}_3)(\text{NH}_2)-(\text{CH}_2)_2-\text{NH}-\text{C}(=\text{O})-\text{CH}_3$	200	5	>95	>95
4.	$\text{CH}_3-\text{C}(=\text{O})-\text{NH}-(\text{CH}_2)_3-\text{C}(\text{CH}_3)(\text{NH}_2)-\text{CH}_2-\text{NH}-\text{C}(=\text{O})-\text{CH}_3$	200	5	>95	>90
5.	$\text{CH}_3-\text{C}(=\text{O})-\text{NH}-\text{CH}_2-\text{C}(\text{CH}_3)(\text{NH}_2)-\text{CH}_2-\text{NH}-\text{C}(=\text{O})-\text{CH}_3$	200	5	>95	NO
6.	$\text{CH}_3\text{CH}_2-\text{C}(=\text{O})-\text{NH}-(\text{CH}_2)_3-\text{NH}-\text{C}(=\text{O})-\text{CH}_2\text{CH}_3$	218	5	>90	>90
7.	$\text{CH}_3-\text{C}(=\text{O})-\text{NH}-(\text{CH}_2)_6-\text{C}(=\text{O})-\text{NH}-\text{CH}_3$	200	5	>95	>90
8.	$(\text{CH}_3)_2-\text{N}-\text{C}(=\text{O})-(\text{CH}_2)_6-\text{C}(=\text{O})-\text{N}-(\text{CH}_3)_2$	228	5	>90	>90
9.	$(\text{CH}_3)_2-\text{N}-\text{C}(=\text{O})-(\text{CH}_2)_7-\text{C}(=\text{O})-\text{N}-(\text{CH}_3)_2$	242	2	>90	>90
10.	$\text{NH}-(\text{CH}_2)_6-\text{NH}-(\text{CH}_2)_6-\text{NH}-\text{C}(=\text{O})-\text{CH}_3$	241	2	>90	>90
11.	$(\text{CH}_3)_2-\text{N}-\text{C}(=\text{O})-(\text{CH}_2)_3-\text{C}(\text{CH}_3)(\text{NH}_2)-\text{C}(=\text{O})-\text{N}-(\text{CH}_3)_2$	269	2.5	>90	>90
12.	$(\text{CH}_3)_2-\text{N}-\text{C}(=\text{O})-(\text{CH}_2)_3-\text{C}(\text{CH}_3)(\text{NH}_2)-\text{C}(=\text{O})-\text{N}-(\text{CH}_3)_2$	442	0.6	>90	>90

Table 2

			24h		48h		120h		commitment	
			cell count	B+	cell count	B+	cell count	B+	24h	48h
HMBA	1 mM	DS19	0.4	0-1	1.0	0-1	3.0 x10 <sup>6</sup>	8%	0-1	6-8
		V <sub>3</sub> 17	0.38	8-9	0.8	33%	2.8	58%	30%	43%
	2 mM	DS19	0.42	0-1	0.9	0-1	2.9	64%	0-1	71%
		V <sub>3</sub> 17	0.4	7-9	0.8	72%	2.6	89%	39%	79%
	3 mM	DS19	0.4	0-1	0.9	2%	2.8	82%	0-1	81%
		V <sub>3</sub> 17	0.3	9-10	0.4	80%	1.8	91%	77%	94%
	4 mM	DS19	0.3	0-1	0.8	11%	2.6	94%	4%	92%
		V <sub>3</sub> 17	0.28	9-13	0.4	88%	1.6	96%	80%	99%
	5 mM	DS19	0.23	0-1	0.5	14%	2.4	98%	6%	96%
		V <sub>3</sub> 17	0.22	9-14	0.5	89%	1.6	99%	83%	98%
IC-135	1 mM	DS19	0.3	0-1	0.6	0-1	1.6	69%	0-1	71%
		V <sub>3</sub> 17	0.3	8-9	0.6	83%	1.3	89%	58%	89%
	2 mM	DS19	0.3	0-1	0.5	0-1	1.3	79%	1-2	87%
		V <sub>3</sub> 17	0.29	7-6	0.4	88%	1.3	94%	70%	95%
	3 mM	DS19	0.27	0-1	0.4	0-1	1.3	86%	5-4	91%
		V <sub>3</sub> 17	0.12	8-6	0.3	89%	0.9	96%	76%	97%
	4 mM	DS19	0.13	0-1	0.2	0-1	0.6	78%	2-1	79%
		V <sub>3</sub> 17	0.18	6-8	0.19	34%	0.5	79%	44%	81%
	5 mM	DS19	0.12	0	0.13	0-1	0.4	--	8%	--
		V <sub>3</sub> 17	0.12	8-11	0.12	0-1	0.3	--	0-1	--
CONTROL			0.4	5-9	1.0	0.1	3.1x10 <sup>6</sup>	0-1	0-1	0-1

IC-135 50 mM Stock. 40  $\lambda$  -- 1mM  
200  $\lambda$  -- 5mM

HMBA 200 mM Stock 10  $\lambda$  -- 1mM  
50  $\lambda$  -- 5mM.

Table 3

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TESTING OF IC-135 ON DIFFERENTMELC LINES

10

DS-1q	1st Day	2nd Day	5th Day Cell Count	5th B <sup>+</sup>	(5 day commitment)
Control	0.38	1.1	2.8	0-1	0-1
5 mM HMBA	0.2	0.5	2.0	91-90	88-88
0.5 mM IC-135	0.38	1	2.1	41-49	31-28
1 mM IC-135	0.33	0.8	1.4	72-74	51-58
2 mM IC-135	0.2	0.5	1.3	80-80	78-76
3 mM IC-135	0.2	0.5	1.2	87-88	81-80

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V3-17	1st Day	2nd Day	5th Day	5th B <sup>+</sup>	(5 day commitment)
Control	0.4	1.2	2.9	3-6	0-1
5 mM HMBA	0.23	0.5	1.6	98-99	99-98
0.5 mM IC-135	0.4	0.6	2.7	40-49	36-34
1 mM IC-135	0.31	0.44	1.3	89-83	89-91
2 mM IC-135	0.21	0.4	1.0	96-91	92-94
3 mM IC-135	0.23	0.3	0.9	98-97	99-99

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30

DR-10	1st Day	2nd Day	5th Day	5th B <sup>+</sup>	5th day commitment)
Control	0.41	1.0	2.9	0-1	0-0
5 mM HMBA	0.3	0.48	2.1	90-89	1-4
0.5 mM IC-135	0.4	0.62	1.9	44-39	0-0
1 mM IC-135	0.32	0.48	1.4	70-74	0-1
2 mM IC-135	0.2	0.41	1.2	79-81	0-2
3 mM IC-135	0.2	0.40	1.2	88-86	1-3

35

Chemistry

5 HMBA, compound 1 of Table 1, (9) was obtained from Aldrich Chemical Co.

10 The preparation and characterization of compounds 1, 2, 6, 7, 8 and 9 of Table 1 (9,27) was previously described. All new amides were purified by chromatography on alumina with 5% methanol in methylene chloride, and were judged pure by thin layer chromatography (single spot) and <sup>1</sup>H NMR spectroscopy. Final products were characterized by <sup>1</sup>H NMR, infrared, and CI mass spectroscopy, while intermediates were characterized by <sup>1</sup>H NMR spectra. The data were consistent with the assigned structures (expected infrared and NMR signals, M + 1 mass spectra).

20 For the synthesis of compound 3 (Table 1), the known 3-methyl-1,5-dibromopentane (38) was converted to the bis-phthalimide derivative, and this was cleaved with hydrazine to afford 3-methyl-1,5-diaminopentane, isolated as the dihydrochloride (m.p. 123-126°). This was converted to compound 3 with acetic anhydride and triethylamine in dioxane.

25 Compound 4 (Table 1) (m.p. 67-68°) was obtained in quantitative yield by similar acetylation of commercially available 2-methyl-1,5-diaminopentane.

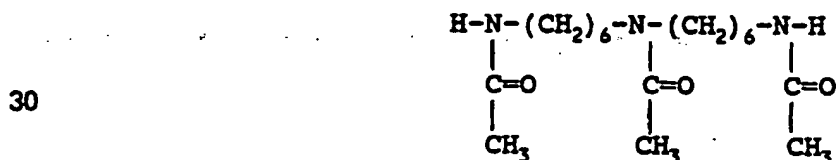
30 Compound 5 (Table 1) was synthesized from meso-2,3-dimethylsuccinic acid in six steps. Reduction of the dimethyl ester with LiAlH<sub>4</sub> afforded the meso-2,3-dimethylbutanediol (92% yield). This was converted to the bis-tosylate, and then to the bis-phthalimide. Deprotection

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and acetylation as before gave compound 5 as an oil (61% yield from the diol).

5 Compound 10 (Table 1) was prepared by making a solution of 19.8gm commercial bis-hexamethylenetriamine in 500 ml of 1,4-dioxane at room temperature under argon. Then 44.8 ml. of triethylamine was added, and 20.3 ml. of acetyl chloride was slowly added with stirring. After two hours of stirring  
10 at room temperature the triethylamine hydrochloride was removed by filtration and the filtrate was concentrated in vacuo. The product triacetyl compound was isolated as a clear viscous oil at about 90% yield by chromatography on basic alumina using isopropanol/ethyl acetate/dichloromethane in the ratio 2/3/5. On thin layer  
15 plates of basic alumina with this solvent mixture the product had an  $R_f$  of ca. 0.6.

20 The mass spectrum (chemical ionization,  $\text{NH}_3$  carrier) showed peaks at 342 (100%,  $M + 1$ ), 227 (10%) and 115 (22%). The infrared spectrum (thin film on NaCl) had bands at 3288, 2931, 2858, 1627, 1560, 1437, 1373, and 1292  $\text{cm}^{-1}$ . In the proton NMR ( $\text{CDCl}_3$ ) the acetyl groups appeared at 6.10 as a broad signal, while the methylene protons appeared as  
25 multiples with the expected intensities in the regions of 3.12 to 3.30 and 1.21 to 1.54.



bis-Hexamethylenetriacetamide (IC-135)

35 For preparation of the triamide compound 11 (Table 1),

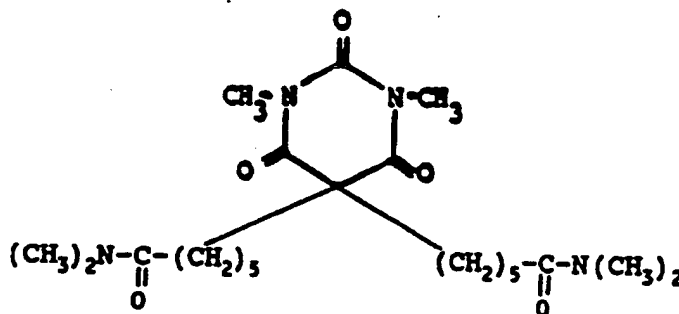


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dimethyl malonate was dialkylated with ethyl 6-bromohexanoate under standard conditions. The resulting tetraester was then hydrolyzed and thermally monodecarboxylated to 1,7,13-tridecanetricarboxylic acid. Treatment with thionyl chloride followed by diethylamine afforded compound 11 (Table 1).

Compound 12 (Table 1) was prepared by simple dialkylation of dimethyl malonate with the known N,N'-dimethyl-6-bromohexanecarboxamide (39).

The compound having the structure:



was made as follows. 1,3-dimethylbarbituric acid (5g; 0.032 mol) in DMF (100 ml) was slowly added to a suspension of sodium hydride (1.5g; 0.064 mol) in DMF (300 ml). The suspension was stirred at 80°C for five hours. N,N-Dimethyl 6-bromohexanoylamide (14.3 g; 0.064 mol) in DMF (100 ml) was added to the cool suspension with stirring. The resulting suspension was stirred at 120°C overnight. The DMF was evaporated and the residue was partitioned between chloroform and water (100-100 ml). The chloroform layer was separated, and the water layer was extracted with chloroform (5 x 50 ml). The combined chloroform layers were dried over anhydrous magnesium sulfate, and evaporated. The oily residue was purified by column chromatography on silica gel in ethyl acetate-tetrahydrofuran (2:1). The yield of 1,3-

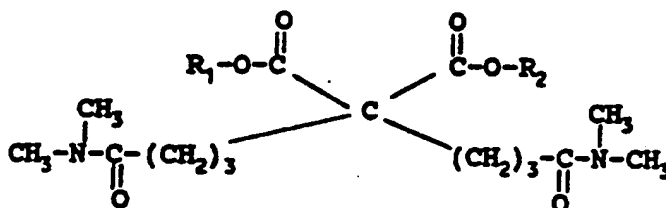
dimethyl-5,5-di(N,N-dimethyl-5-pentylcaboxamide)barbituric acid was 4.1 g (30%).

- 5 <sup>1</sup>H NMR (CDCl<sub>3</sub>, 200 MHz) δ 3.33 (s, bar, N-CH<sub>3</sub>, 6H), 2.98 (s, N-CH<sub>3</sub>, 6H), 2.93 (s, N-CH<sub>3</sub>, 6H), 2.24 (t, J=7 Hz, CH<sub>2</sub>CON, 4H), 1.96 (m, 4H), 1.58 (m, 4H), 1.28 (m, 4H), 1.08 (m, 4H).

The compound having the structure:

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- was made as follows. Diethyl malonate (50.6 ml; 53.3 g; 0.33 mol) in DMF (150 ml) was added slowly to a cool suspension of sodium hydride (16 g; 0.67 mol) in DMF (1 liter). The suspension was carefully heated with stirring at 80° C (about one hour). The clear mixture was cooled, and N,N-dimethyl 6-bromohexanoylamide (146.5 g; 0.66 mol) in DMF (200 ml) was slowly added at room temperature. The suspension was stirred at 110°C for two hours, and the DMF was evaporated. The semisolid residue was partitioned between chloroform and water (about 200-200 ml). The organic layer was separated, and the water layer was extracted with chloroform (5 x 100 ml). The combined chloroform layers were dried over anhydrous magnesium sulfated and evaporated. The oily residue was purified by column chromatography on silica gel in ethyl acetate-tetrahydrofuran (4:1). The yield of diethyl undecane-6,6-di(ethylcarboxylate)-1,11-di(N,N-dimethylcarboxamide) was

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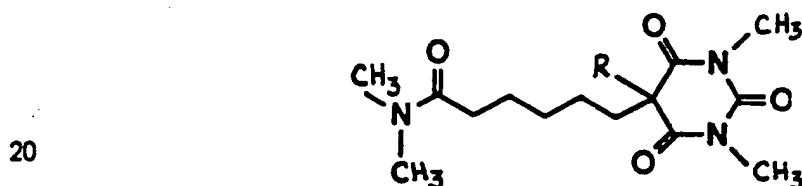
106 g (76%).

<sup>1</sup>H-NMR(CDCl<sub>3</sub>, 200 MHz) δ4.11(q, J=7.0 Hz, 4H), 2.96  
5 (t, J=7.5Hz, 1H), 2.89(s, 6H), 2.25(t, J=7.2Hz, 4H), 1.88 (m,  
4H), 1.59 (m, 4H), 1.16-1.31 (m, 14H).

IR (NaCl, CH<sub>2</sub>Cl<sub>2</sub>) 2937, 2866, 1728, 1642, 1498, 1462, 1399,  
10 1369, 1265, 1153, 1097, 1029 cm<sup>-1</sup>.

MS (FAB, glycerol) m/e=443 (40%, M+1<sup>+</sup>), 398 (25%), 295  
(80%), 142 (45%).

15 The compound having the structure:

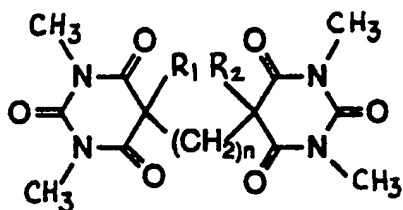


25 was made as follows. 1,3,5,-trimethylbarbituric acid (1.7g;  
10 mmol) in DMF (20ml) was added at room temperature to a  
suspension of sodium hydride (0.24 g; 10 mmol) in DMF (100  
ml). The suspension was stirred at 90°C for one hour. N,N-  
Dimethyl 6-bromohexanoylamide (2.22 g; 10 mmol) in DMF (20  
ml) was added to a cool mixture (about 5°C). The suspension  
30 was stirred at 110°C overnight. DMF was evaporated, and the  
residue was partitioned between chloroform and water (50-50  
ml). The chloroform layer was separated, and the water  
layer was extracted with chloroform. The combined  
chloroform layers were dried over anhydrous magnesium  
35 sulfate and evaporated. The oily residue was purified by

column chromatography on silica gel in ethyl acetate-tetrahydrofuran. The yield of 1,3-dimethyl-5-methyl-5-(5-pentyl-N,N-dimethylcarboxamide)barbituric acid was 1.5 g (48%).

$^1\text{H}$  NMR ( $\text{CDCl}_3$ , 200 MHz)  $\delta$  3.23 (s, bar.H, N-CH<sub>3</sub>, 6H), 2.99 (s, N-CH<sub>3</sub>, 3H), 2.94 (s, N-CH<sub>3</sub>, 3H), 2.26 (t, J=7.2 Hz, CH<sub>2</sub>CON, 2H), 1.98 (m, 2H), 1.62 (m, 2H), 1.52 (m, C-Me, 3H), 1.32 (m, 2H), 1.11 (m, 2H).

The compounds having the structure:

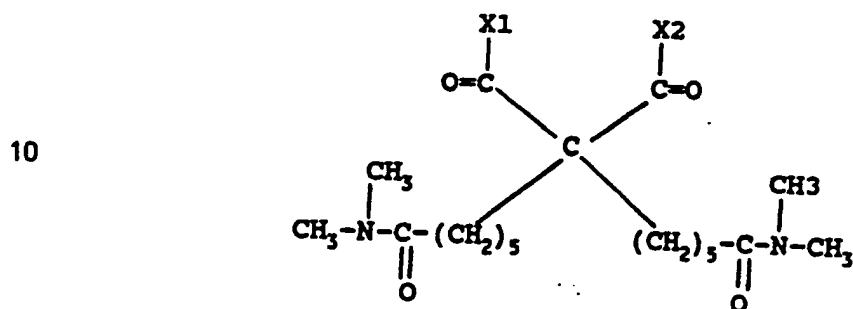


were made as follows. 1,3,5-trimethylbarbituric acid (0.71 g; 0.029 mol) in DMF (10 ml) was added to a suspension of sodium hydride (0.71g; 0.029 mol) in DMF (20 ml). The suspension was stirred at 80°C for one hour. 1,n-dibromoalkane (0.0147 mol) was added to a cool suspension (about 5°C). The reaction mixture was heated at 110°C for four hours, cooled down and evaporated. The solid residue was slurried in water (about 200 ml), filtered, and the white solid was washed with water (3 x 5 ml), ethanol (3 x 20 ml), and ether (2 x 20 ml). The yield of dibarbituric acids were for: n=3, 75%; n=4, 78%; n=5, 71%; n=6, 79%; n=7, 83%.

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$^1\text{H}$  NMR for the compound with  $n=2$  ( $\text{CDCl}_3$ , 200 MHz)  $\delta$ 3.32(s, N-CH<sub>3</sub>, 12H), 1.86 (s, CH<sub>2</sub>, 4H), 1.47 (s, C-CH<sub>3</sub>, 6H).

5 The compounds having the structure:



15 were made as follows. When X1 and X2 are the same and each is  $\text{HNCH}_3$ , aqueous (10 ml) potassium hydroxide (1.53 g; 27.5 mmol) was added to the diamidebarbituric acid (2.4 g; 5.5 mmol) in water (50 ml). The mixture was stirred at 70° C for fifteen minutes, cooled, filtered, acidified, and  
20 extracted by chloroform (5 x 20 ml). The combined chloroform extracts were dried over anhydrous magnesium sulfate and evaporated to the oily residue. By the NMR, hexamethylamide was more than 96% pure. The yield of undecane-1,11-di-(N,N-dimethylcarboxamide)-6,6-di(N-methylcarboxamide) was 2.1 g (93%).  
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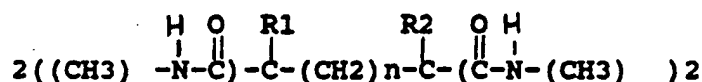
$^1\text{H}$  NMR ( $\text{CDCl}_3$ , 200 MHz)  $\delta$ 7.38 (q,  $J=8$  Hz), 3.03 (s, N-CH<sub>3</sub>, 6H), 2.94 (s, N-CH<sub>3</sub>, 6H), 2.8 (d,  $J=8$  Hz, N-CH<sub>3</sub>, 6H), 2.32 (t,  $J=7.2$  Hz, CH<sub>2</sub>CON, 4H), 2.82 (m, 4H), 2.62 (m, 4H), 1.25 (m, 8H).  
30

When X1 and X2 are the same and each is  $\text{N}(\text{CH}_3)_2$ , the hexamethylamide (2 g; 4.9 mol) in DMF (10 ml) was added at room temperature to a suspension of sodium hydride (0.24 g;  
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0.01 mol) in DMF (20 ml). The mixture was stirred at 80°C for one hour. Methyl iodide (3.5 ml; 8 g; 56 mmol) was added with stirring to the cool reaction mixture. The reaction was continued at 60°C overnight. The solvent was evaporated and the solid residue was dissolved in water (50 ml) and the water solution was extracted with 8:2 chloroform-methanol (5 x 50 ml). The combined organic layers were dissolved in acetone (200 ml) and filtered through a short column of silica gel. Acetone was evaporated yielding pure undecane-1,6,6,11-tetra-(N,N-dimethylcarboxamide) (1.36 g; 63%).

<sup>1</sup>H NMR (CDCl<sub>3</sub>) δ2.99 (s, N-CH<sub>3</sub>, 6H), 2.94 (s, N-CH<sub>3</sub>, 6H), 2.92 (s, N-CH<sub>3</sub>, 6H), 2.88 (s, N-CH<sub>3</sub>, 6H), 2.28 (t, J=7.2 Hz, CH<sub>2</sub>CON, 4H), 1.86 (m, 4H), 1.62 (m, 4H), 1.32 (m, 8H).

The compounds having the structure:



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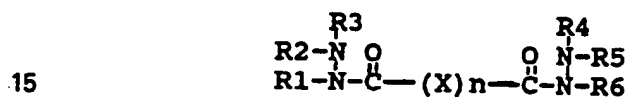
wherein n is an integer from 1 to 15 are made as follows. A dibarbituric acid (3.66 g; 0.01 mol for n = 2) was suspended in 9:1 water-dioxane (60 ml) and heated up to 100°C. To this suspension was added at 110°C a solution of potassium hydroxide (5.6 g; 0.1 mol/20 ml water for n=2). The suspension immediately became a solution. The solution was heated at 110°C for an extra 10 minutes, cooled down, filtered, and acidified. The clear water solution was extracted with chloroform (10 x 50 ml). The combined

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chloroform extracts were dried over anhydrous magnesium sulfate, and evaporated. The yields of the alkane-1,1,n,n,-tetra-(N-methylcarboxamide) were for: n=2, 85%; n=3, 79%; n=4, 83%; n=5, 81%; n=6, 74%; n=7, 83%.

<sup>1</sup>H NMR (CDCl<sub>3</sub>, 200 MHz) for compounds having n=2, δ7.92 (q, J=4.6 Hz, NH-Me, 4H), 2.85 (d, J=4.6 Hz, N-CH<sub>3</sub>, 12H), 1.85 (s, CH<sub>2</sub>, 4H), 1.43 (s, C-CH<sub>3</sub>, 6H).

The compounds having the structure:



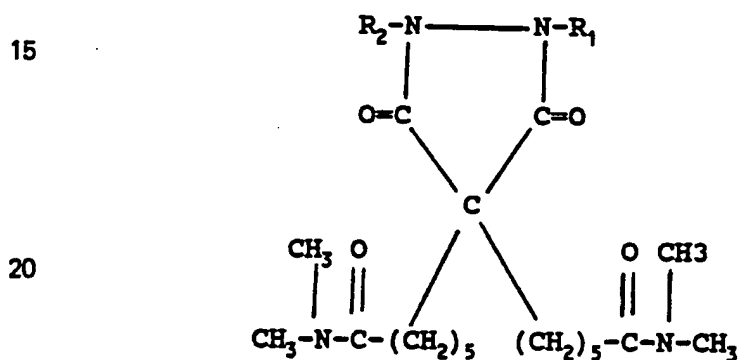
wherein R<sub>1</sub>, R<sub>2</sub>, R<sub>3</sub>, R<sub>4</sub>, R<sub>5</sub>, R<sub>6</sub> may independently be the same or different from each other and is H or a lower alkyl group; wherein X is methyl or phenyl; and n is an integer from 1 to about 15, were made as follows. When R<sub>1</sub> and R<sub>6</sub> is each H and R<sub>2</sub>, R<sub>3</sub>, R<sub>4</sub>, and R<sub>5</sub> is each CH<sub>3</sub>, to a chloroform (200 ml) solution of suberoyl chloride (8 g; 38 mmol) was added at room temperature to 1,1-dimethylhydrazine (11.5 ml; 9.1 g; 152 mmol). The mixture was stirred at room temperature for about three hours. The solvent was evaporated, and the residue was dissolved in methanol (400 ml). The methanol was evaporated and the solid residue was dissolved in water (100 ml). The water solution was extracted with hexanes (5 x 50 ml), and with chloroform (5 x 50 ml). The combined extracts were dried over anhydrous magnesium sulfate and evaporated. The solid residue was crystallized from acetone. The yield of hexane-1,6-di(N<sup>2</sup>,N<sup>2</sup>-dimethyl-carboxhydrazide) was 6.5 g (66%).

When R<sub>1</sub>, R<sub>2</sub>, R<sub>3</sub>, R<sub>4</sub>, R<sub>5</sub>, and R<sub>6</sub> is each CH<sub>3</sub>, to a suspension

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of sodium hydride ( 1.4 g; 57.9 mmol) in DMF (60 ml) at room temperature was added tetramethylhydrazide (5 g; 19.3 mmol) followed by methyl iodide (31.5 ml; 71.8 g; 579 mmol). The reaction mixture was stirred at 60°C for five hours. The DMF was evaporated, and the solid residue was dissolved in water (50 ml) and extracted with chloroform (5 x 50 ml). The chloroform extracts were dried over anhydrous magnesium sulfate and evaporated. The product was purified by crystallization from hexanes. The yield of hexane-1,6-di(N<sup>1</sup>,N<sup>2</sup>,2-trimethylcarboxhydrazide) was 3.7 g (67 %).

The compounds having the structure:

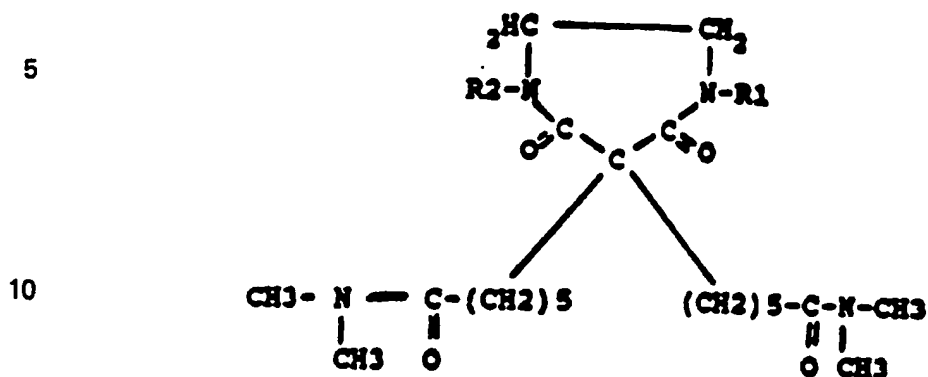


may be made as follows. The heterocyclic compounds may be alkylated after they are formed from malonic chloride and N,N'-dialkylhydrazine.



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The compounds having the structure:

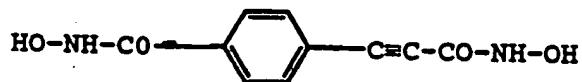


wherein R<sub>1</sub> and R<sub>2</sub> may be the same or different and is H or CH<sub>3</sub>, may be made as follows. The heterocyclic part of the compound may be made from malonic chloride and N,N'-dealkylethylenediamine. The heterocyclic compound may then be alkylated with the corresponding alkyl bromide under known conditions.

20 The compounds having the structure:



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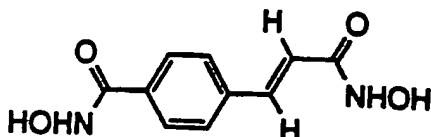


30 may be synthesized from the corresponding acid via acidic chloride with hydroxylamine hydrochloride in a solution of tetrahydrofuran and water. The acid may be made from 1,4-benzenediacrylic acid by an addition of bromine, and following elimination with base.

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The compound having the structure:

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10 was made as follows. A suspension of 4-carboxycinnamic acid (10 g; 0.052 mol) and thionyl chloride (15.2 ml), 24.8 g; 0.208 mol) in benzene (500 ml) was refluxed for five hours. The solvent was evaporated and the oily residue was dissolved in tetrahydrofuran (100 ml). The tetrahydrofuran solution was slowly added at about 5°C to a stirred water solution (100 ml) of hydroxylamine hydrochloride (7.3 g; 0.104 mol) and potassium hydroxide (11.6 g; 0.208 mol). A white precipitate was formed immediately. The suspension was stirred at room temperature for two hours, and filtered. 15 The solid was washed with cooled water (5 x 20 ml) and acetone (5 x 20 ml). The solid was slurried in methanol (3 liters), and the suspension was refluxed. The hot suspension was filtered. The filtrate was concentrated to 100 ml and cooled. The white crystals were separated by filtration, and washed with acetone (3 x 20 ml) and ether (5 x 20 ml). The yield of benzene-1-(N-hydroxycarboxamide)-4-[trans-ethene-2-(N-hydroxycarboxamide)] was 5.2 g (45%).

20 <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 200 MHz), 11.05 (broad s, OH, 2H), 9.12 (broad s, NH, 2H), 7.77 (d, J=8.2 Hz, arom. H, 2H), 7.62 (d, J=8.2 Hz, arom. H, 2H), 7.47 (d, J=15.8 Hz, CH, 2H), 6.53 (d, J=15.8 Hz, CH, 2H).

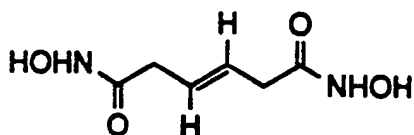
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The compound having the structure:

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15 was made as follows. A suspension of trans-B-hydromuconic acid (5 g; 0.0347 mol), oxalyl chloride (12 ml; 17.6 g; 0.139 mol) in benzene (250 ml) and a few drops of DMF was stirred at room temperature until the reaction suspension became clear (about one hour). The solvent was evaporated, and the oily residue was dissolved in tetrahydrofuran (50 ml). The tetrahydrofuran solution was slowly added into cool (about 5°C) aqueous solution of hydroxylamine hydrochloride (4.8 g; 0.0694 mol) and potassium hydroxide (7.8 g; 0.1388 mol). The reaction suspension was stirred at room temperature for about one hour and evaporated to a volume of 20 ml. The suspension was cooled and filtered. The crude solid product was slurried in methanol (200 ml) and the suspension was cooled and filtered. The filtrate was dried over anhydrous magnesium sulfate, and evaporated to a solid residue. The solid was slurried in ether (250 ml) and recovered by filtration. The yield of trans-2-butene-1,4-di-(N-hydroxylcarboxamide) was 3.8 g (63%).

35 <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 200 MHz), δ 10.85 (s, OH, 2H), 9.08 (s, NH, 2H), 7.11 (d of d, J<sup>1</sup>=11.2 Hz, J<sup>2</sup>=2.8 Hz, CH, 2H), 6.17 (d

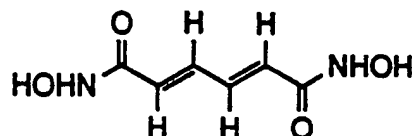
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of d,  $J^1=11.4$  Hz,  $J^2=2.8$  Hz).

The compound having the structure:

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was made as follows. A solution of trans, trans-muconic acid (5 g; 0.035 mol), oxalyl chloride (12.3 ml, 17.9 g; 0.141 mol) and a few drops N,N-dimethyl formamide in benzene (250 ml) was stirred at room temperature for three hours. The solvent was removed under reduced pressure. The oily residue was dissolved in tetrahydrofuran (50 ml) and the tetrahydrofuran solution was added dropwise to the stirred solution of hydroxylamine hydrochloride (4.9 g; 0.074 mol) and potassium hydroxide (7.9 g; 0.141 mol) at about 5°C. A white precipitate was formed immediately. The suspension was stirred at room temperature for about one hour and filtered. The solid was slurried in methanol (2 liters) and the suspension then was boiled and filtered. The methanol solution was concentrated to a volume of about 20 ml. The solid product was separated by filtration and washed with ether (5 x 20 ml). The yield of trans,trans-1,3-butadiene-1,4-(N-hydroxylcarboxamide) was 3.1 g (52%).

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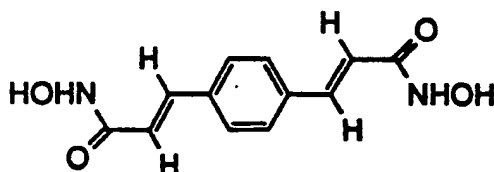
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$^1\text{H}$  NMR ( $\text{DMSO}-d_6$ , 200 MHz), 10.46 (s, 2H, OH), 8.75 (s, 2H, NH), 5.53 (t,  $J=3.4$  Hz, CH, 2H), 2.70 (d,  $J=3.4$  Hz, 4H,  $\text{CH}_2$ ).

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The compound having the structure:

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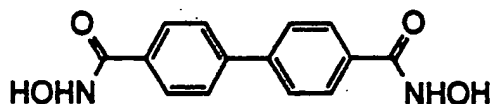
20 was made as follows. A suspension of 1,4-phenylenediacrylic acid (10 g; 0.184 mol) and thionyl chloride (13.4 ml; 21.9 g; 0.184 mol) in benzene (500 ml) was refluxed for five hours. The clear solution was cooled, and the solvent was evaporated. The oily residue was dissolved in  
25 tetrahydrofuran (100 ml) and slowly added into cooled (about 5°C) aqueous (100 ml) solution of hydroxylamine hydrochloride (6.4 g; 0.092 mol) and potassium hydroxide (10.3 g; 0.184 mol). A white precipitate was formed immediately. The suspension was stirred at room temperature  
30 (two hours), and filtered. The crude product was slurried in boiling water (100 ml), and the suspension was filtered. The solid was slurried in acetone (200 ml) and boiled. The hot suspension was filtered, and the solid was washed with acetone (5 x 20 ml), and then with ether (5 x 20 ml). The  
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yield of benzene-1,4-di[trans-ethene-2-(N-hydroxulcarboxamide)] was 6.2 g (54%).

5  $^1\text{H}$  NMR (DMSO- $d_6$ , 200 Hz),  $\delta$  10.82 (s, OH, 2H), 9.17 (s, NH, 2H), 7.58 (s,  $\text{C}_6\text{H}_4$ , 4H), 7.45 (d,  $J=15.8$  Hz, CH, 4H), 6.51 (d,  $J=15.8$  Hz).

The compound having the structure:



was made as follows. A suspension of 4,4-diphenyldicarboxylic acid (5 g; 20.6 mmol) and thionyl chloride (7.3 ml; 12 g; 0.1 mmol) in benzene (250 ml) was refluxed overnight. The clear solution was cooled, and the solvent was evaporated. The solid residue was dissolved in tetrahydrofuran (100 ml) and was added slowly into a cool (about 5°C) aqueous (100 ml) solution of hydroxylamine hydrochloride (2.9 g; 41.7 mmol) and potassium hydroxide (4.65 g; 83 mmol). A white precipitate was formed immediately. The suspension was stirred at room temperature for about two hours and filtered. The crude product was slurried in water (50 ml) and the resulting suspension was boiled (about 10 minutes). The solid was separated by filtration, and slurried in acetone (200 ml) and boiled. The hot suspension was filtered, and the solid was washed with acetone (5 x 50 ml), and then with ether (5 x 20 ml). The yield of diphenyl-4,4'-di-(N-hydroxylcarbox-amide) was

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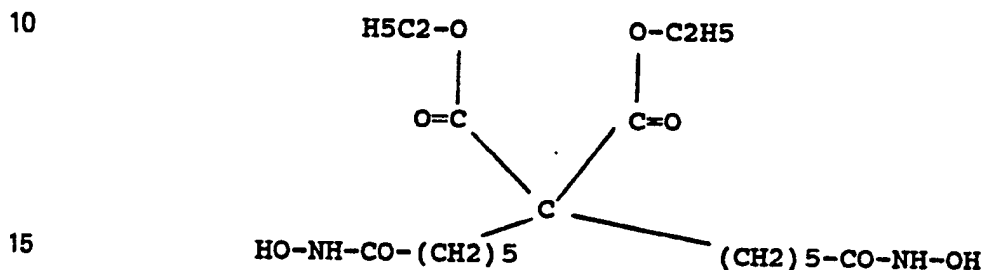
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4.9 g (88%).

<sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 200 MHz), δ 11.33 (s, OH, 2H), 9.12 (broad s, NH, 2H), 7.87 (d, J=8.2 Hz, arom. H, 2H), 7.80 (d, J=8.4 Hz, arom. H, 2H).

The compound having the structure:



was made as follows. Diethyl malonate (15.2 ml; 16 g; 0.1 mol) was slowly added at room temperature to a suspension of sodium hydride (4.8 g; 0.2 mol) in N,N-dimethyl formamide (300 ml). The suspension was heated at 80°C for about one hour and cooled down. Benzyl-6-bromohexanoate (57 g; 0.2 mol) was added at room temperature with vigorous stirring. The suspension was heated at 100°C for about 5 hours and the DMF was evaporated. The residue was partitioned between chloroform and water (200-200 ml). The chloroform layer was separated and the water layer was extracted with chloroform (5 x 50 ml). The combined chloroform layers were dried over anhydrous magnesium sulfate, and evaporated. The oily residue was purified by column chromatography on silica gel in hexanes-ethyl acetate. The yield of undecane-6,6-di(ethylcarboxylate)-1,11-di(benzylcarboxylate) was 41 g (72 %). Dibenzyl 6,6-di(ethoxycarbonyl)-1,11-undecanedicarboxylate (20 g; 35.2 mmol) was dissolved in methanol (150 ml) and 5 % Pd-C (200 mg) was added. The reaction

suspension was hydrogenated at room temperature overnight. The catalyst was removed by filtration, and the methanol was evaporated yielding pure 6,6-(ethylcarboxylate)-1,11-undecanedicarboxylic acid (12.5 g; 93%).

The acid was suspended in benzene (500 ml), and oxalyl chloride (11.2 ml; 16.3 g; 0.129 mol) and a few drops of DMF were added. The mixture was stirred at room temperature for about 3 hours. The solvent was evaporated and the oily residue was dissolved in chloroform (100 ml). The chloroform solution of the acid chloride was slowly added to a stirred solution of O-benzylhydroxylamine (16.3 g; 0.132 mol) in chloroform (300 ml). A white precipitate was formed immediately. The suspension was filtered, and concentrated hydrochloric acid was added (50 ml) to the chloroform filtrate and again filtered. The chloroform layer was washed with concentrated hydrochloric acid (3 x 100 ml), water (3 x 100 ml), and dried over anhydrous magnesium sulfate. The chloroform was evaporated yielding pure N,N'-dibenzyl-6,6-di(ethoxycarbonyl)-1,11-undecanedicarboxamide (17.1 g; 89%).

The benzyloxyamide (6 g; 0.01 mol) was dissolved in methanol (50 ml), and 5 % Pd-C (200 mg) was added. The methanol suspension was hydrogenated at room temperature overnight. The catalyst was separated by filtration and methanol was evaporated yielding pure undecane-6,6-di(ethylcarboxylate)-1,11-di(N-hydroxylcarboxamide) (3.9 g; 93%). The overall yield was 55.4%.

$^1\text{H}$  NMR (DMSO- $d_6$ , 200 MHz), 10.30 (broad s, OH, 2H), 8.75 (broad s, NH, 2H), 4.10 (q,  $J=7$  Hz,  $\text{COOCH}_2\text{CH}_3$ , 4H), 1.90 (t,  $J=7.2$  Hz,  $\text{CH}_2\text{CONHOH}$ , 4H), 1.71 (m, 4H), 1.45 (m, 4H), 1.18



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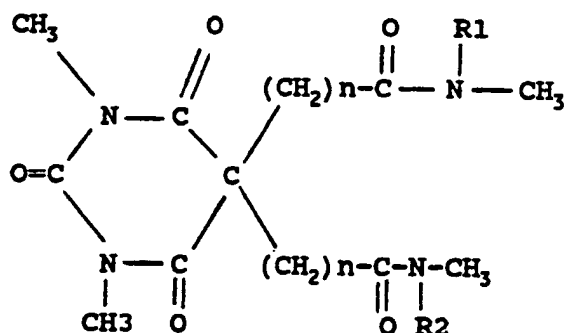
(m, 8H), 1.14 (t, J=7 Hz, COOCH<sub>2</sub>CH<sub>3</sub>, 6H).

The compounds having the structure:

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wherein R1 and R1 may independently be the same or different and is H or a lower alkyl group, wherein R3 and R4 may independently be the same or different and each is CH<sub>3</sub> or OH, and wherein n is 5 or 6 were made as follows. 1,3-dimethylbarbituric acid (15.6 g; 0.1 mol) was slowly added at room temperature to a suspension of sodium hydride (4.8 g; 0.2 mol) in N,N-dimethylformamide (300 ml). The reaction was heated with stirring at 100°C for about three hours. A DMF (200 ml) solution of benzyl 6-bromohexanoate (57 g; 0.2 mol) was slowly added to the cool suspension with vigorous stirring. The suspension was heated at 110°C overnight, and the solvent was evaporated. The residue was partitioned between chloroform and water (300-300 ml). The chloroform layer was separated and the water layer was extracted with chloroform (5 x 50 ml). The combined chloroform layers were dried over anhydrous magnesium sulfate, and evaporated. The oily residue was purified by column chromatography on silica gel in ethyl acetate-tetrahydrofuran (9:1). The yield of

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dibenzyl ester was 18 g (32%).

5 The dibenzyl ester (18 g; 0.032 mol) was dissolved in methanol (50 ml) and catalyst (200mg of 5 % Pd-C) was added. The suspension was hydrogenated overnight, and the catalyst was removed by filtration. The methanol was evaporated yielding pure acid (11.5 g; 94%). The acid (11 g; 0.0286 mol) was suspended in benzene (500 ml), and oxalyl chloride (10ml; 14.5 g; 0.14 mol) and a few drops of DMF were added. 10 The mixture was stirred at room temperature for about 3 hours and evaporated to an oily residue. The oily residue was dissolved in chloroform (100ml) and poured into a chloroform (100 ml) solution of O-benzyl-hydroxylamine (14.4 g; 0.117 mol). A white precipitate was formed immediately. 15 After two hours stirring at room temperature the suspension was filtered. Concentrated hydrochloric acid (50 ml) was added to the filtrate and again filtered. The chloroform layer was separated and washed with concentrated 20 hydrochloric acid (3 x 100 ml), water (3 x 100 ml) and dried over anhydrous magnesium sulfate. The chloroform was evaporated yielding pure N-benzyloxyamide (14 g; 82%).

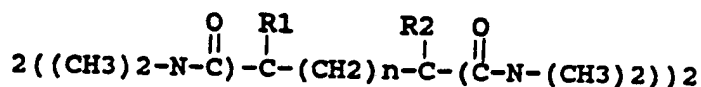
25 The N-benzyloxyamide (6 g; 0.01 mol) was dissolved in methanol (509 ml) and catalyst (200 mg of 5% Pd-C) was added. The suspension was hydrogenated at room temperature overnight. The catalyst was removed by filtration and the methanol was evaporated yielding pure hydroxamic acid (3.95 g; 95%). The overall yield of 1,3-dimethyl-5,5-di(5-pentyl- 30 n-hydroxylcaboxamide)barbituric acid was 23.4%.

35 <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 200 MHz), δ 10.30 (s, OH, 2H), 8.60 (broad s, NH, 2H), 3.18 (s, N-CH<sub>3</sub>, 6H), 1.85 (t, J=7.2 Hz,

57

CH<sub>2</sub>CONHOH, 4H), 1.80 (t, J=7 Hz, CH<sub>2</sub>-bar., 4H), 1.39 (m, 4H), 1.10 (m, 8H).

5 The compound having the structure:

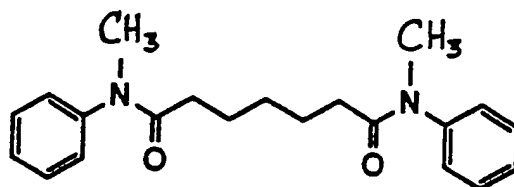


10 wherein R1 and R2 independently be the same or different and each may is H or a lower alkyl group and n is an integer from 1 to 10 is synthesized as follows. The tetramethylamide (2 g; 6.4 mmol; for n=2) in DMF (10 ml) was slowly added to a suspension of sodium hydride (0.9; 38  
15 mmol) in DMF (50 ml). The mixture was stirred at 80°C for about 30 minutes. Iodomethane (23.3 ml; 53.2 g; 380 mmol) was added to this cool solution at a temperature of about 5°C and the reaction was continued at 60°C overnight. The solvent was evaporated, and the residue was dissolved in  
20 water (50 ml). The water solution was extracted with 20% methanol in chloroform (10 x 30 ml). The combined organic extracts were dried over anhydrous magnesium sulfate and evaporated. The product was purified by short column chromatography in acetone. The yields of the alkane-  
25 1,1,n,n-tetra-(N,N-dimethylcarboxamide) were for: n=2, 1.3 g (55%); n=3 (52%); n=4, 51%; n=5 (53%); n=6 (53%).

<sup>1</sup>H NMR (CDCl<sub>3</sub>, 200 MHz) for compound having n=2, δ 2.94 (s, N-Me, 12 H), 2.87 (s, N-Me, 12H), 1.82 (s, CH<sub>2</sub>, 4H), 1.32 (s, C-Me, 6H).  
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58

The compound having the structure:



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was synthesized as follows. N-methylanyline (7.3 ml; 7.219 g; 67.3 mmol) was slowly added to a solution of pimeloyl chloride (5 ml; 6.025 g; 30.6 mmol) in chloroform (500 ml). The suspension was stirred at room temperature for about three hours, and the solid was separated by filtration. The chloroform filtrate was washed with concentrated hydrochloric acid (3 x 100 ml), water (3 x 100 ml) and dried over anhydrous magnesium sulfate. The chloroform was evaporated, the solid residue was slurried in hexanes and filtered. The yield of pentane-1,5-di(N-methyl-N-phenylcarboxamide) was 6.3 g (90%).

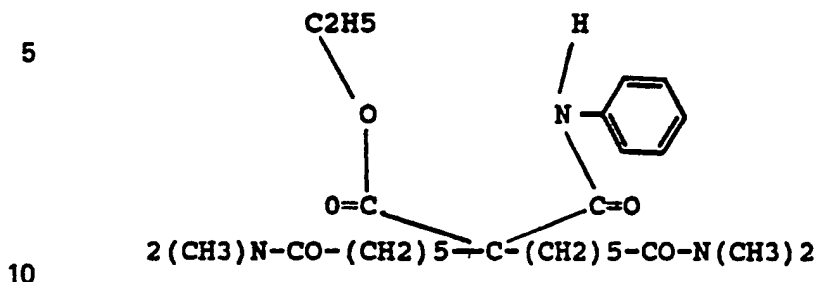
<sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 200 MHz), δ 7.30 (m, Ph, 10 H), 3.12 (s, N-Me, 6H), 1.93 (t, J=7 Hz, CH<sub>2</sub>CON, 4H), 3.31 (m, 4H), 1.01 (m, 2H).

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The compound having the structure:

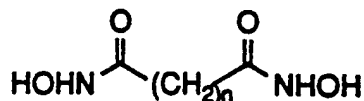


was synthesized as follows. To a suspension of sodium hydride (0.24 g; 0.01 mol) in DMF (30ml) at room temperature was added slowly to aniline (1 ml; 1 g; 10.7 mmol) in DMF (5 ml). The mixture was heated at 110 °C for three hours. Diethyl di(N,N-dimethylpentamethylenecarboxamide)malonate (4.42 g; 0.01 mol) in DMF (10 ml) was added to the cool reaction mixture. The mixture was stirred at 110°C overnight. The solvent was evaporated. The residue was partitioned between water and chloroform (100-100 ml). The chloroform layer was separated, and the water layer was extracted with 4:1 chloroform-methanol (5 x 50 ml). The combined organic layers were dried over anhydrous magnesium sulfate and evaporated to an oily residue. The product was purified by column chromatography on silica gel in tetrahydrofuran. The yield of undecane-1,11-di(N,N-dimethyl-carboxamide)-6-ethoxycarbonyl-6-(N-phenylcarboxamide) was 2.1 g (43%).

<sup>1</sup>H NMR (CDCl<sub>3</sub>, 200 MHz), δ 7.59 (d, J=7.4 Hz, arom. 2H), 7.28 (t, J=7.4 Hz, arom. 2H), 7.26 (s, NH, 1H), 7.10 (t, J=7.4 Hz, arom. 1H), 4.26 (q, J=7 Hz, COOCH<sub>2</sub>CH<sub>3</sub>, 2H), 2.99 (s, N-Me, 6H), 2.95 (s, N-Me, 6H), 2.24 (t, J=5.2 Hz, CH<sub>2</sub>CON, 4H) 1.85 (m, 4H), 1.60 (m, 4H), 1.65 (m, 8H).

60

The compounds having the structure:



10 were made as follows. A solution of suberoyl chloride (n=6; 10 ml; 11.72 g; 55.5 mmol) in tetrahydrofuran (50 ml) was slowly added (about 1 hour) to a stirred solution of potassium hydroxide (12.4 g; 22.2 mmol) and hydroxylamine hydrochloride (7.72 g; 111 mmol) in water (200 ml), maintaining the reaction temperature below 5°C. The

15 reaction suspension was stirred an additional two hours at room temperature. The white precipitate (where n= 6-11) was filtered and the filtrate was washed with cold water (5 x 20 ml), acetone (5 x 20 ml) and ether (3 x 50 ml). The product was more than 95% pure by NMR(DMSO). The product can be

20 purified by crystallization from acetone. When n=2-5, the reaction mixture was evaporated. The white crystals were slurried in hot methanol (about 1 liter) and filtered. The filtrate was dried over anhydrous magnesium sulfate and the methanol was evaporated. The white crystals were purified

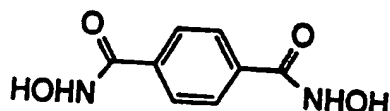
25 by crystallization from methanol or acetone. The yields of alkane-1,n-di(N-hydroxylcarboxamide) were for n=2, 57%; n=3, 64%; n=5, 71%; n=6, 63%; n=7, 68%; n=8, 78%; n=9, 75%; n=10; 79%; and n=1, 81%.

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The compound having the structure:



10 was synthesized as follows. A solution of tetraphthaloyl  
chloride (5 g; 24 mmol) in tetrahydrofuran (50 ml) was  
slowly added (about 30 minutes) to a stirred solution of  
potassium hydroxide (5.52 g; 98.6 mmol) and hydroxylamine  
hydrochloride (3.4 g; 48.9 mmol) in water (100 ml) and the  
15 reaction was maintained at a temperature below 5°C. The  
reaction suspension was stirred an additional hour at room  
temperature. The suspension was evaporated, and the solid  
was slurried in boiling methanol (about 1 liter). The hot  
methanol suspension was filtered, the filtrate was dried  
20 over anhydrous magnesium sulfate and the methanol was  
evaporated. The solid residue was slurried in ethyl acetate  
(30 ml). The solid was separated by filtration and washed  
with ether (5 x 10 ml) and hexanes (5 x 10 ml). The yield  
of benzene-1,4-di(N-hydroxylcarboxamide) was 3.2 g; (66%).  
25

The compound having the structure:



30 was made by slowly adding a solution of suberoyl chloride (5  
g; 23.7 mmol) in tetrahydrofuran (20 ml) to a stirred water  
solution of potassium hydroxide (5.3 g; 94.8 mmol) and N-  
methylhydroxylamine hydrochloride (4 g; 47.4 mmol) at about  
5°C. The reaction mixture was stirred at room temperature  
35

for about one hour. The organic layer was separated and the water layer was extracted with chloroform (5 x 50 ml). The combined chloroform layers were dried over anhydrous magnesium sulfate and evaporated. The product was purified by crystallization from acetone. The yield of hexane-1,6-di(N-hydroxyl-N-methylcarboxamide was 2.7 g (46%).

Each compound in Table 1 was assayed 3 or more times to determine effectiveness as an inducer of MELC (745A-DS19) cell line. The cell density of MELC in culture for 5 d. without inducer was  $2.0$  to  $2.6 \times 10^6$  cells/ml. The cell density of MELC in culture for 5 d. with inducer was  $1.2$  to  $2.0 \times 10^6$  cells/ml. Compounds 11 and 12 were dissolved in absolute ethyl alcohol. The final concentration of ethyl alcohol in the cultured medium ranged between 0.1 and 3%. This concentration of ethyl alcohol had no effect on cell growth of MELC in culture without inducer. All other compounds were dissolved in culture medium without fetal calf serum. The indicated optimal concentration represents the final concentration in the culture medium.

## Results

### New Hybrid Polar/Apolar Compounds Active as Cytodifferentiation Agents

In earlier studies (8), the present inventors reported that fairly high concentrations of certain polar organic solvents induce MELC to undergo erythroid differentiation. The question arose as to how the effectiveness of polar compounds might be increased so that similar concentrations of these solvent-like molecules could induce



differentiation. Although the mechanisms were not clear, and are still incompletely understood, the following hypothesis was considered: perhaps, at the target site of action, more than one solvent molecule must bind or interact. if this were so, the well-known chelate effect could provide more effective compounds. Instead of binding two or more independent solvent molecules, the cellular target might interact more efficiently with a single molecule carrying two or more solvent-like groups. Provided these groups were held in the right relationship one to the other, binding of one would carry the other into the target region providing a high, effective concentration. Such a chelate effect is well precedented; it accounts for the strong binding of metal ions by chelating ligands, and is the basis for much of the catalytic activity of enzyme.

This concept led the present inventors to new effective cytodifferentiating agents. The best studied to date is hexamethylene bisacetamide (HMBA, Table 1, compound 1), consisting of two acetamide molecules linked at nitrogen by a six carbon polyethylene chain (5,9). N-methyl acetamide (Table 1, compound 2), another of the polar organic solvents, was shown to be effective, but only at high concentration (50 mM), whereas HMBA induces erythroid differentiation in MELC at an optimal concentration of 5 mM (8). The present inventors previously showed that the optimum number of methylene groups in the apolar chain is six (27). In the present invention, it was found that a four or five carbon chain is comparably effective if extra carbons are provided as branching methyl groups (Table 1, compounds 3, 4 and 5). This suggests that the important factor is not simply the length of the chain but also the number of hydrophobic hydrocarbon units in it.

Acetamide can also be dimerized by linking the methyl groups of the molecule. Suberic acid bis-N-methylamide (SBDA; Table 1, compound 7), can be thought of as N-methylacetamide  
5 linked at the acetyl group by four methylene units. SBDA is comparable in activity as an inducer to HMBA (Figure 1A and B). Since the structures of SBDA and HMBA are different, it is likely that the metabolic fates of these two compounds are completely different, and their similarity in  
10 effectiveness means that the compounds themselves are the principal active agents, rather than their catabolic products.

HMBA and SBDA have their polar groups separated by identical  
15 methylene bridges, and they have a similar ratio of polar to apolar hydrophobic moieties. Many structurally related compounds have been examined, but only a few showed comparable or greater activity (Table 1, compounds 6 through 12). It is clear that the structures of active compounds  
20 may differ sufficiently from HMBA to make it likely that they will also display different pharmacokinetics. One of the more active of these hybrid compounds is a dimer of HMBA. Bisexamethylene triacetamide (BHTA; compound 10) is about 2 fold more active as inducer, on a molar basis, than  
25 HMBA (Figure 1C). The most active of the hybrid compounds assayed in this study is one in which two pentamethylene carboxyamides are linked by the dimethyl ester of malonic acid (compound 12). This compound, with 4 exposed polar groups balanced by two apolar pentamethylene domains, is  
30 roughly 10 fold more active than HMBA. For example, 0.6 mM compound 12 is about as effective as 5.0 mM HMBA, inducing over 90% of cells to differentiate after 5 d. in culture (Fig. 2).

Polymethylenediamine derivatives carrying propionyl groups instead of the acetyl groups of HMBA are also active, but methoxycarbonyl groups are less effective and bulky pivaloyl groups lead to loss of activity. The present inventors previously showed that HMBA can have a double bond (cis or trans) or a triple bond in the center and retain its activity (27). Replacement of the polymethylene chain with a cyclohexane ring leads to inactivity (27), although compound 2, with a longer chain interrupted by an amide group, is active.

Diamides of dicarboxylic acids, such as suberic acid, are active with either one or two methyl groups on each nitrogen (compounds 7, 8, and 9), but not with a methyl and a methoxyl substituent, and they are also active with one (but not two) ethyl groups on each nitrogen. Suberic diamides of pyrrolidine, of morpholine, or of piperazine are inactive. This shows that there is a limit to the amount of carbon tolerated on the ends of the polar groups.

These findings indicate that for optimal activity, two, or even better, three or four uncharged polar groups of limited bulk must be connected by chains of about 5 to 6 carbons.

#### Increased Sensitivity of Vincristine-Resistant MELC to Polar/Apolar Compounds

As previously reported, MELC resistant to relatively low concentrations of vincristine show a marked increase in sensitivity to HMBA (Fig. 1A) (26). The dose-response of MELC to several of the new polar compounds identified as being as or more active than HMBA was examined. The compounds assayed for inducing activity with vincristine-

resistant MELC include compounds 1, 3, 4, 8, 10, 11 and 12 (Table 1). In each instance, vincristine-resistant MELC were induced at lower concentrations than were required for induction of vincristine-resistant MELC were induced more rapidly than the vincristine-sensitive cells. For the most active of the compounds, 0.1 mM compound 12 was the optimal concentration for inducing vincristine-resistant MELC (VCR.C(2)15). For example, 0.1 mM compound 12 induced only 6% commitment of over 95% of V.C.R.(2) 15 cells after 2 d. and the accumulation of over 80% benzidine-sensitive (DS-19) MELC, 0.1 mM compound 12 induced only 6% commitment by d. 2 and 4% benzidine reactive cells by d. 5. At concentrations of compound 12 as low as 0.05, 35% of VCR.C(2)15 became benzidine reactive by d. 2, compared to only 2% of DS-19.

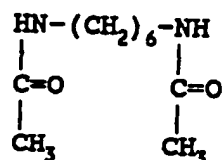
Effect of Polar/Apolar Compounds on MELC Resistant to Inducer Mediated Terminal Cell Division

Several of the new polar/apolar compounds were evaluated as inducers of MELC cell line DR10, which is resistant to induction by dimethylsulfoxide and can be induced by HMBA to accumulate hemoglobin but not commit to terminal cell division (37). Compounds 7, 8 and 10, assayed as inducers of DR10, caused accumulation of hemaglobin, but not commitment to terminal cell division. Thus, these new hybrid polar/apolar compounds are similar to HMBA in their effect on DR10 cells.

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Cell cultures were grown in the presence of different concentrations of Hexamethylenebisacetamide (HMBA)

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and compound 10, bis-Hexamethylenetriacetamide (IC-135).

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At 1, 2 and 5 days, the cell densities and the percentage of cells that were benzidine reactive (B+) were measured. Table 2 shows the cell densities, B+%, and percent of cells committed for cell lines DS19 and V3.17 grown in 1 mM to 5 mM of HMBA and IC-135. Figures 3, 4, 5A and B, and 6A and B are graphical representations of the data presented in Table 2.

15

Table 3 shows cell counts for days 1, 2 and 5 and percentage of cells committed and benzidine reactive (B+) at day 5 for cell lines DS-19, V3.17 and DR-10 grown in 5 mM of HMBA and 0.5 to 3 mM of IC-135.

20

As can be seen in Table 2 and 3 and Figures 3-6, IC-135 is more reactive in the tested cell lines at lower concentrations than HMBA.

25

Table 4 shows the results of several structurally different compounds when assayed 3 or more times to determine effectiveness as an inducer of MELC (745A-DS19) cell line. The cell density of MELC in culture for 5 d. without inducer was 2.0 to 2.6 x 10<sup>6</sup> cells/ml.

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### Discussion

5 The development of agents which can induce transformed cells to differentiate and suppress their oncogenicity has important implications for the treatment of cancer. While a number of agents have been identified that can induce tumor cells in vitro to express features of their differentiated phenotype and to decrease their proliferative capacity (4,10-24), these agents have generally proved to be 10 relatively ineffective or too toxic when evaluated in clinical trials (40).

15 Among cytodifferentiation agents, the hybrid polar/apolar compound, HMBA, has been one of the best characterized with respect to its in vitro inducing activity in MELC and in a number of other transformed cell lines, as well as for certain human tumor explants (30). It can trigger a differentiation program in transformed cells which is 20 similar to that of their normal lineage (5).

The recent finding that vincristine-resistant MELC are more sensitive to HMBA and to other active hybrid polar/apolar compounds provides an approach toward identifying the 25 mechanisms of HMBA action (26). The lack of a latent period for inducer-mediated differentiation of vincristine-resistant cells suggests that an early effect of inducers may involve alternations (e.g. new proteins, or modification of existing proteins such as phosphorylation) which are 30 constitutively expressed in the vincristine-resistant cell lines, and may, therefore, be identified and characterized.

The observed positive therapeutic responses to HMBA, albeit largely transient, occurred despite relatively low serum 35

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concentrations of HMBA (<1 mM), compared to the optimum demonstrated in vitro (4 to 5 mM) (35,36). The present invention provides a new group of hybrid polar/apolar compounds which are as active or even more active, on a molar basis, than HMBA.

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Table 4

		Mol.	Optional	
	<u>Structure</u>	<u>Weight</u>	<u>Concen-</u>	<u>Benzidine</u>
			<u>tration</u>	<u>Reactive</u>
			<u>(μM)</u>	<u>Cells (%)</u>
5				
	$\begin{array}{c} \text{H O} \qquad \qquad \text{O H} \\   \quad    \qquad \quad    \quad   \\ \text{HO-N-C-(CH}_2\text{)}_n\text{-C-N-OH} \end{array}$			
10				
	13. n=2	148	0	
	14. n=3	162	0	
	15. n=4	176	300	60
15	16. n=5	190	40	89
	17. n=6			
	18. n=7	218	20	75
	19. n=8	232	20	38
20	20. n=9	246	20	20
	21. n=10	260	20	26
	22. n=11	274	20	9



Table 4 (continued)


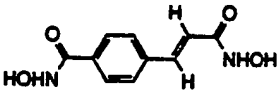
5	<u>CPD</u>	<u>Structure</u>	Mol. <u>Weight</u>	Optional Concen- tration <u>(<math>\mu</math>M)</u>	Benzidine Reactive <u>Cells (%)</u>
10	23.		196	80	88
15					
20	24.			5	64
25					
30					
35					

Table 4 (continued)

5	<u>CPD</u>	<u>Structure</u>	Mol.	Optional Concen- tration ( $\mu$ M)	Benzidine Reactive Cells (%)	Commit- ment (%)
			<u>Weight</u>			
10	25.		438	600	95%	
20	26.		424	500	19%	

73

5	<u>CPD</u>	<u>Structure</u>	<u>Mol.</u> <u>Weight</u>	<u>Optional</u> <u>Concen-</u> <u>tration</u> <u>(<math>\mu</math>M)</u>	<u>Benzidine</u> <u>Reactive</u> <u>Cells (%)</u>
27.					
10					
15			311	5000	85
28.					
20					
25			414	600	85
29.					
30					
35			442.60	600	95

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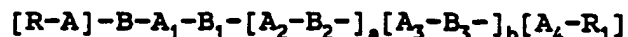
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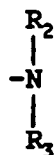
What is claimed is:

1. A compound having the structure:



wherein each A, A<sub>1</sub>, A<sub>2</sub>, A<sub>3</sub>, and A<sub>4</sub> represent a polar group which comprises a nitrogen, sulfur or oxygen atom and wherein each of A, A<sub>1</sub>, A<sub>2</sub>, A<sub>3</sub>, and A<sub>4</sub> may independently be the same as, or different from, the others of A, A<sub>1</sub>, A<sub>2</sub>, A<sub>3</sub>, and A<sub>4</sub>;

wherein each of R and R<sub>1</sub> is a hydrogen atom; a lower alkyl, alkenyl, or alkynyl group; or a group having the structure:



wherein each of R<sub>2</sub> and R<sub>3</sub> being a hydrogen atom or a lower alkyl, alkenyl, or alkynyl group; and wherein each of R, R<sub>1</sub>, R<sub>2</sub> and R<sub>3</sub> may independently be the same as, or different from, the others of R, R<sub>1</sub>, R<sub>2</sub> and R<sub>3</sub>;

wherein each of [R-A] and [A<sub>4</sub>-R<sub>1</sub>] have a dipolar moment greater than about 2.7 Debye units;

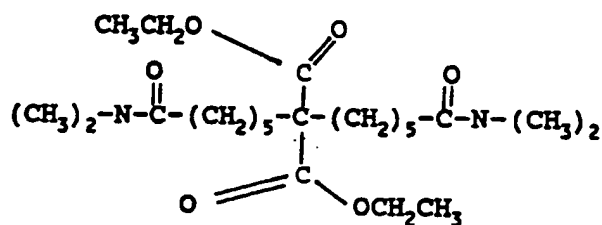
wherein each of B, B<sub>1</sub>, B<sub>2</sub>, and B<sub>3</sub> represents a nonpolar group which comprises at least 4 atoms in a chain, the termini of which chains are attached to A and A<sub>1</sub>, A<sub>1</sub> and A<sub>2</sub>, A<sub>2</sub> and A<sub>3</sub>, and A<sub>3</sub> and A<sub>4</sub>, respectively; wherein each such atom is oxygen, nitrogen, carbon, or sulfur

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and wherein each of B, B<sub>1</sub>, B<sub>2</sub>, and B<sub>3</sub> may independently be the same or different from, each of B, B<sub>1</sub>, B<sub>2</sub>, and B<sub>3</sub>;

and wherein each of a and b is independently 0 or 1.

2. A compound of claim 1 having the structure:



3. A compound of claim 1, wherein each of A, A<sub>1</sub>, A<sub>2</sub>, A<sub>3</sub> and A<sub>4</sub> is a carbonyl radical or a bivalent radical of an amide, a sulfoxide, or an amine oxide.

4. A compound of claim 1, wherein each of A, A<sub>1</sub>, A<sub>2</sub>, A<sub>3</sub> and A<sub>4</sub> is a carbonyl radical or a bivalent radical of an amide.

5. A compound of claim 4, wherein all of A, A<sub>1</sub>, A<sub>2</sub>, A<sub>3</sub>, and A<sub>4</sub> are the same.

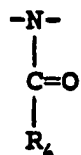
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6. A compound of claim 5, wherein all of A, A<sub>1</sub>, A<sub>2</sub>, A<sub>3</sub> and A<sub>4</sub> have the structure:

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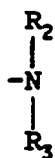
wherein R<sub>4</sub> is a hydrogen atom or a lower alkyl or alkenyl group.

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7. A compound of claim 4, wherein all of A, A<sub>1</sub>, A<sub>2</sub>, A<sub>3</sub> and A<sub>4</sub> are a carbonyl radicals.

8. A compound of claim 1, wherein R and R<sub>1</sub> are the same and are hydrogen atoms, methyl groups, ethyl groups or groups having the structure:

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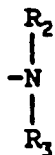
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each of R<sub>2</sub> and R<sub>3</sub> each is hydrogen atom, a methyl group an ethyl group and being the same or different.

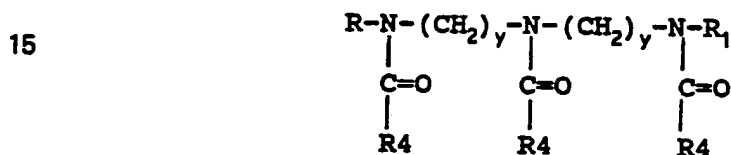
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9. A compound of claim 8, wherein all of A, A<sub>1</sub>, A<sub>2</sub>, A<sub>3</sub>, and A<sub>4</sub> are carbonyl radicals and R and R<sub>1</sub> are:

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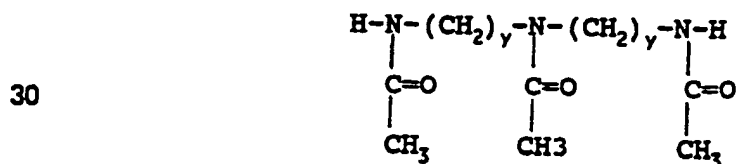
10. A compound of claim 1, wherein each of B, B<sub>1</sub>, B<sub>2</sub>, and B<sub>3</sub> further comprises a nonpolar substituent attached to an atom of a chain.
11. A compound of claim 1, wherein each of B, B<sub>1</sub>, B<sub>2</sub>, and B<sub>3</sub> comprises 4 to 7 atoms connected in a chain.
12. A compound of claim 1, wherein each of B, B<sub>1</sub>, B<sub>2</sub>, and B<sub>3</sub> may comprise a hydrocarbon chain.
13. A compound of claim 1 having the structure:



wherein each of R, R<sub>1</sub>, and R<sub>4</sub> is a hydrogen atom or a methyl, ethyl, or propyl group, R and R<sub>1</sub> are the same or different; and wherein y is 4, 5, or 7.

14. A compound of claim 13, wherein R and R<sub>1</sub> are hydrogen atoms, each R<sub>4</sub> is an ethyl group, and y is 4, 5 or 6.

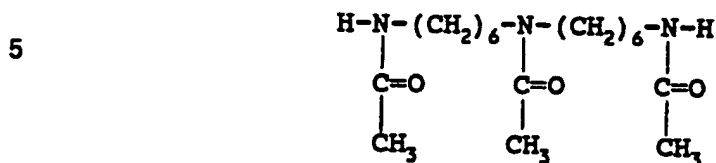
15. A compound of claim 13 having the structure:



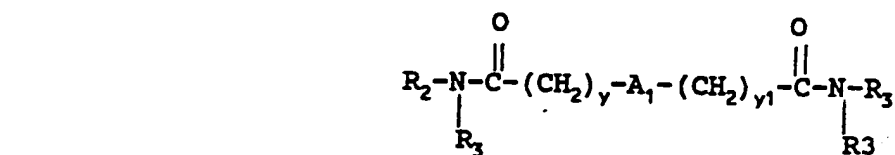
wherein y is 4, 5 or 6.

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16. A compound of claim 15 having the structure:



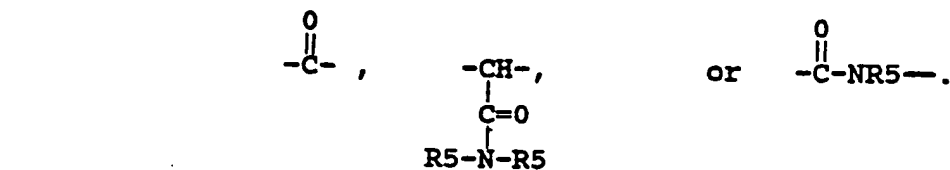
- 10 17. A compound of claim 1, having the structure:



- wherein each of R<sub>2</sub> and R<sub>3</sub> is a hydrogen atom or a methyl, ethyl or propyl group and is the same or different, and wherein each of y and y<sub>1</sub> is independently 4, 5, 6 or 7.

- 20 18. A compound of claim 17, wherein R<sub>2</sub> is a hydrogen atom, R<sub>3</sub> is an ethyl group and each of y and y<sub>1</sub> is independently 4, 5 or 6.

- 25 19. A compound of claim 17, wherein A<sub>1</sub> is

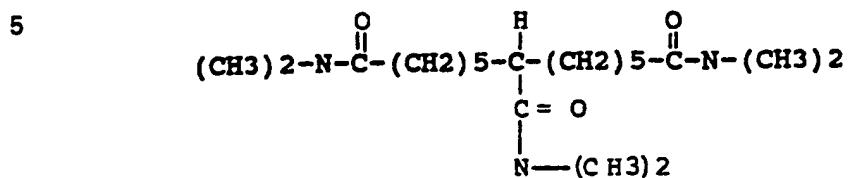


- wherein R<sub>5</sub> is a hydrogen atom or a methyl, ethyl or propyl group.

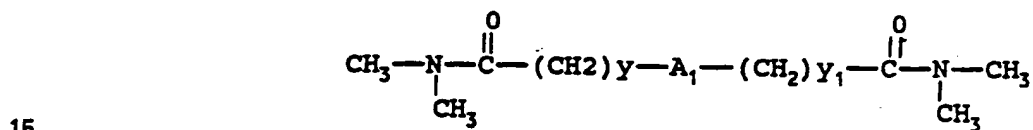
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20. A compound of claim 19 having the structure:

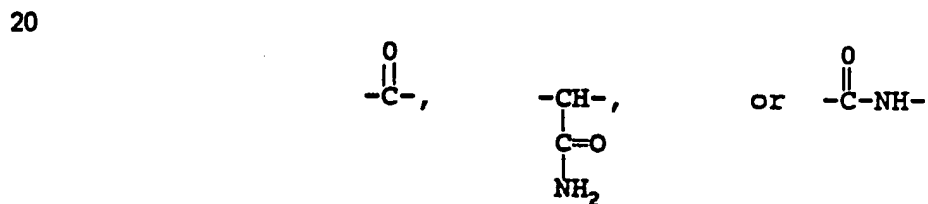


10 21. A compound of claim 18 having the structure:

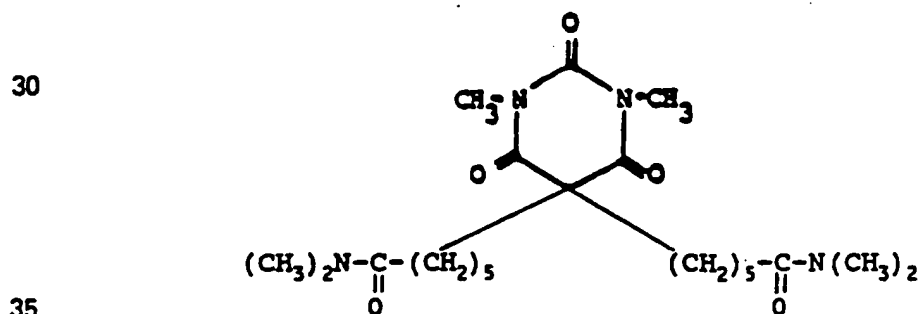


wherein each of  $y$  and  $y_1$  is independently 5 or 6.

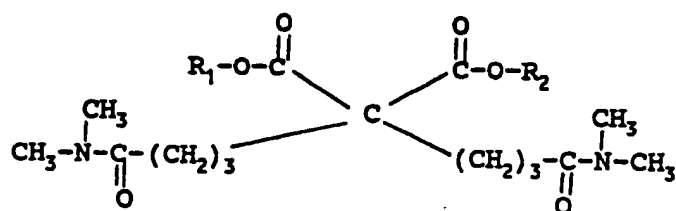
22. A compound of claim 21, wherein  $A_1$  is:



25 23. A compound having the structure:

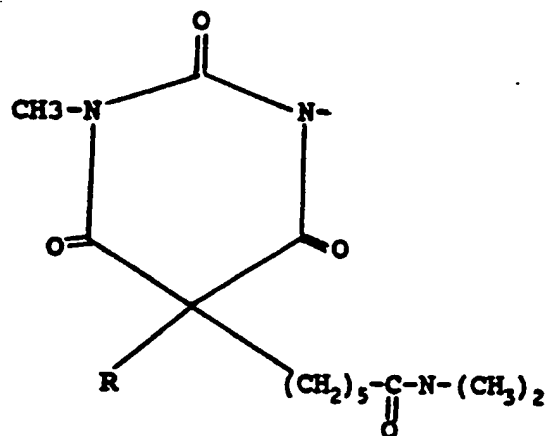


24. A compound having the structure:



wherein R1 and R2 may be the same or different and each is a lower alkyl group.

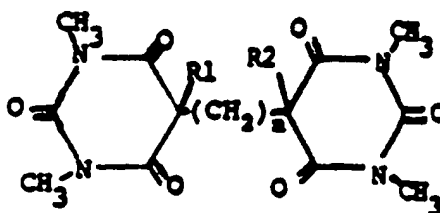
25. A compound having the structure:



wherein R is H or a lower alkyl group.

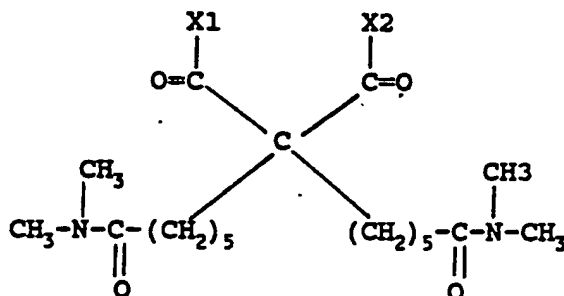
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26. A compound having the structure:



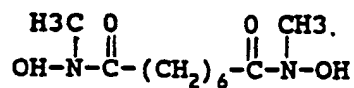
wherein  $n$  is an integer which is greater than 1 and  $R_1$  and  $R_2$  may be the same or different and each is H or a lower alkyl group.

27. A compound having the structure:



wherein  $X_1$  and  $X_2$  may independently be the same or different and each is  $N(CH_3)_2$ ,  $NH$ -phenyl, or  $HNCH_3$ .

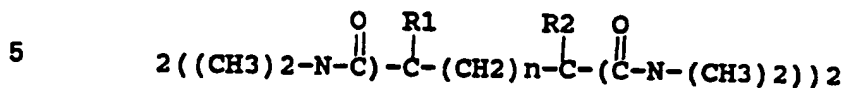
28. A compound having the structure:





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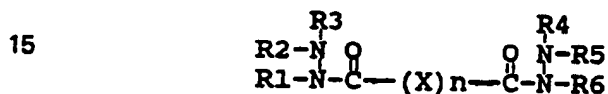
29. A compound having the structure:



wherein R1 and R2 may be the same or different and each is H or a lower alkyl group;

and n is an integer from 1 to 10.

30. A compound having the structure:

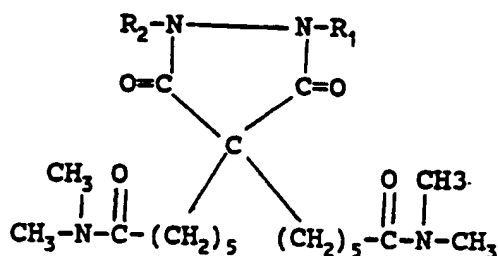


wherein R1, R2, R3, R4, R5, R6 may independently be the same or different from each other and is H or a lower alkyl group;

wherein X is methyl or phenyl; and

n is an integer from 1 to about 15.

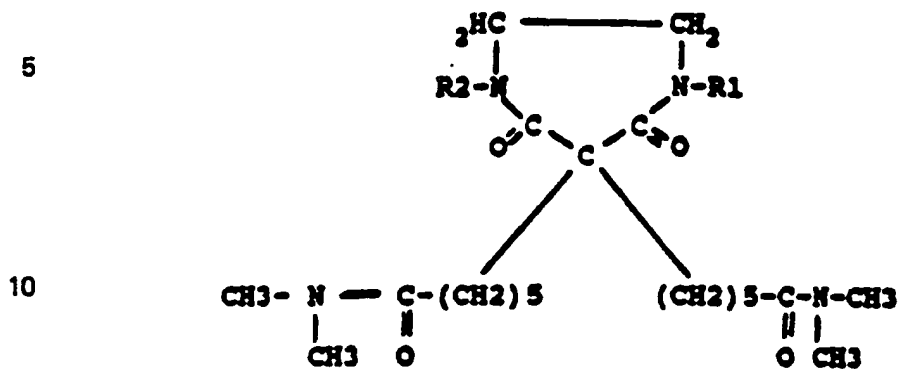
31. A compound having the structure:



wherein R1 and R2 may be the same or different and is H or CH3.

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32. A compound having the structure:

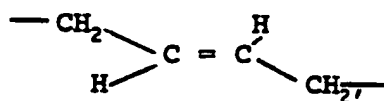
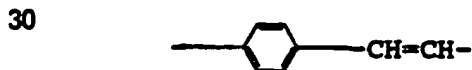
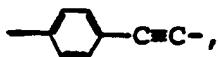
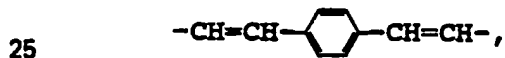


wherein R<sub>1</sub> and R<sub>2</sub> may be the same or different and is H or CH<sub>3</sub>.

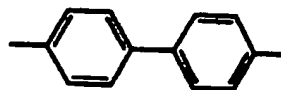
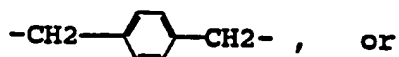
33. A compound having the structure:



wherein X has the structure:



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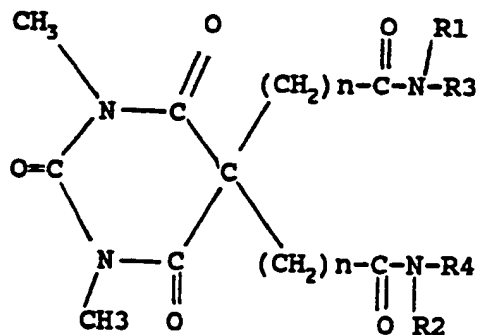


wherein R<sub>1</sub> and R<sub>2</sub> are the same or different and is H or lower alkyl group.

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34. A compound having the structure:

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wherein R<sub>1</sub> and R<sub>1</sub> may independently be the same or different and each may be H or a lower alkyl group;

wherein R<sub>3</sub> and R<sub>4</sub> may independently be the same or different and each may be CH<sub>3</sub> or OH; and

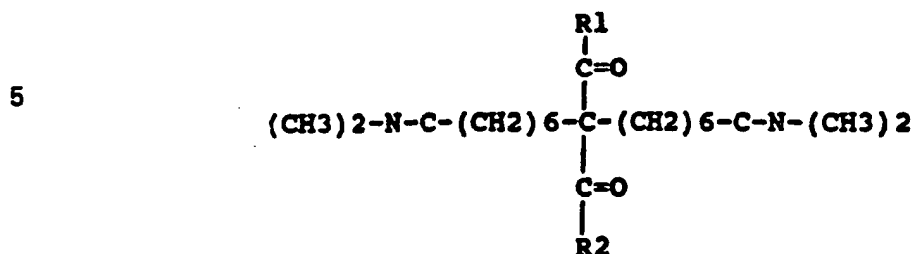
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wherein n is 5 or 6.

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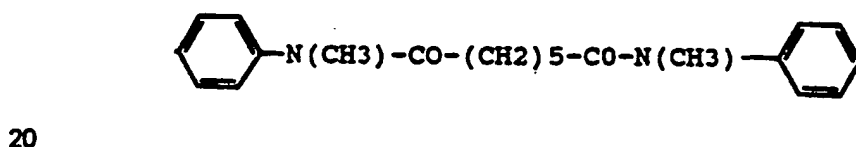
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35. A compound having the structure:

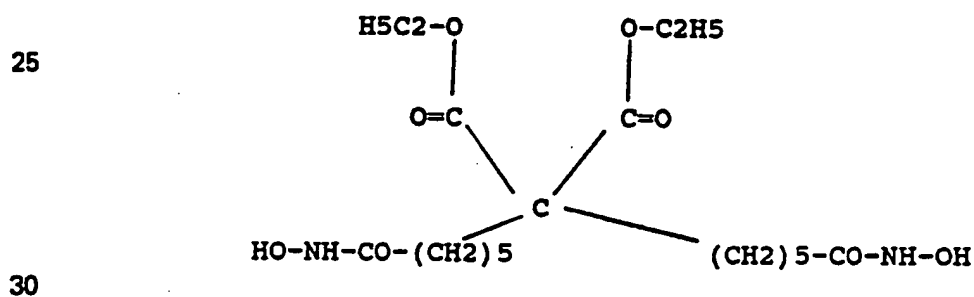


10 wherein  $R_1$  and  $R_2$  is the same or different and is hydrogen, lower alkyl, alkenyl, alkynyl, an amide or hydroxyamide.

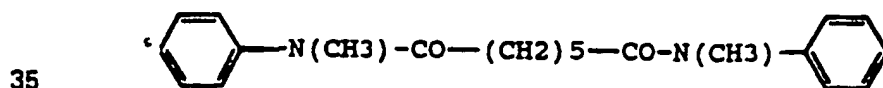
15 36. A compound having the structure:



37. A compound having the structure:

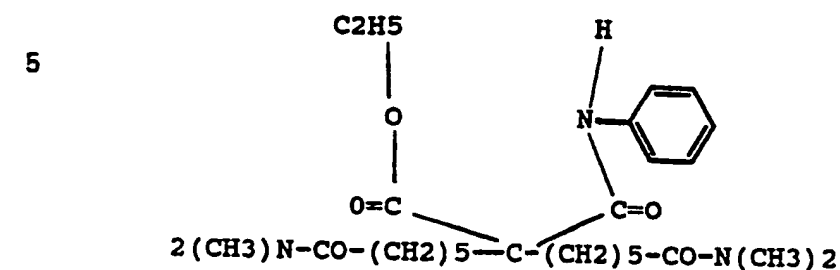


38. A compound having the structure:

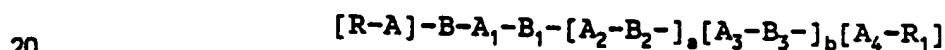


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39. A compound having the structure:

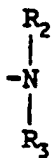


40. A method of selectively inducing terminal differentiation of neoplastic cells and thereby inhibiting proliferation of such cells which comprises contacting the cells under suitable conditions with an amount of a compound effective to selectively induce terminal differentiation, the compound having the structure:



25 wherein each A, A<sub>1</sub>, A<sub>2</sub>, A<sub>3</sub>, and A<sub>4</sub> represent a polar group which comprises a nitrogen, sulfur or oxygen atom and wherein each of A, A<sub>1</sub>, A<sub>2</sub>, A<sub>3</sub>, and A<sub>4</sub> may independently be the same as, or different from, the others of A, A<sub>1</sub>, A<sub>2</sub>, A<sub>3</sub>, and A<sub>4</sub>;

30 wherein each of R and R<sub>1</sub> is a hydrogen atom; a lower alkyl, alkenyl, or alkynyl group; or a group having the structure:



35 wherein each of R<sub>2</sub> and R<sub>3</sub> being a hydrogen atom or a

lower alkyl, alkenyl, or alkynyl group; and wherein each of R, R<sub>1</sub>, R<sub>2</sub> and R<sub>3</sub> may independently be the same as, or different from, the others of R, R<sub>1</sub>, R<sub>2</sub> and R<sub>3</sub>;

5

wherein each of [R-A] and [A<sub>4</sub>-R<sub>1</sub>] have a dipolar moment greater than about 2.7 Debye units;

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wherein each of B, B<sub>1</sub>, B<sub>2</sub>, and B<sub>3</sub> represents a nonpolar group which comprises at least 4 atoms in a chain, the termini of which chains are attached to A and A<sub>1</sub>, A<sub>1</sub> and A<sub>2</sub>, A<sub>2</sub> and A<sub>3</sub>, and A<sub>3</sub> and A<sub>4</sub>, respectively; wherein each such atom is oxygen, nitrogen, carbon, or sulfur and wherein each of B, B<sub>1</sub>, B<sub>2</sub>, and B<sub>3</sub> may independently be the same or different from each of B, B<sub>1</sub>, B<sub>2</sub>, and B<sub>3</sub>;

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and wherein each of a and b is independently 0 or 1.

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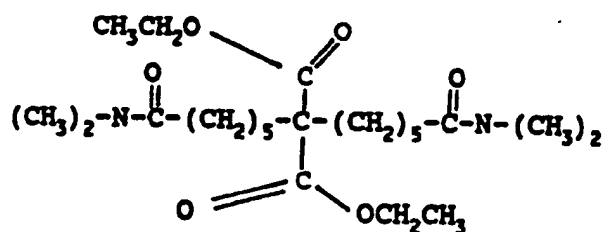
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41. A method of claim 40, wherein the compound has the structure:

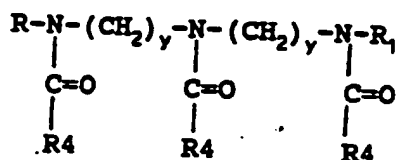
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42. A method of claim 40, wherein the contacting is effected continuously for at least 48 hours.

- 15 43. A method of claim 40, wherein the compound has the structure:



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wherein each of R, R<sub>1</sub>, and R<sub>4</sub> is a hydrogen atom or a methyl, ethyl or propyl group, R and R<sub>1</sub> are the same and R and R<sub>4</sub> are the same or different; and wherein y is 4, 5, 6 or 7.

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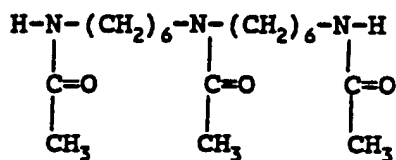
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44. A method of claim 43, wherein the compound has the structure:

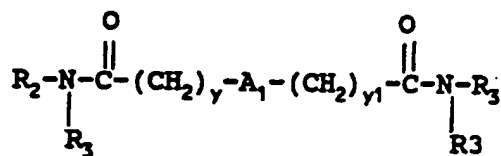
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45. A method of claim 43, wherein the amount of the compound is from about 0.5mM to about 25mM.

46. A method of claim 40, wherein the compound has the structure:

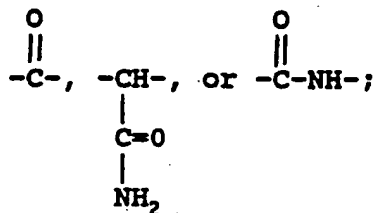
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wherein each of  $\text{R}_2$  and  $\text{R}_3$  is a hydrogen atom or a methyl, ethyl or propyl group and each being the same or different; wherein  $\text{A}_1$  is

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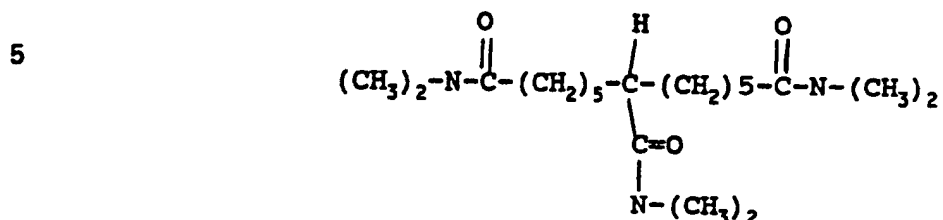
and wherein each of  $y$  and  $y_1$  is independently 5, 6 or 7.

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47. A method of claim 46, wherein the compound has the structure:



- 10 48. A method of claim 47, wherein the amount of the compound is from about 0.1 to about 10 mM.
- 15 49. A method of selectively inducing terminal differentiation of neoplastic cells and thereby inhibiting proliferation of such cells which comprises contacting the cells under suitable conditions with an amount of a compound of claims 23-39, effective to selectively induce terminal differentiation.
- 20 50. A method of selectively inducing terminal differentiation of neoplastic cells and thereby inhibiting proliferation of such cells which comprises contacting the cells under suitable conditions with an amount of a compound effective to selectively induce terminal differentiation, the compound having the structure:
- 25



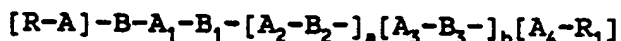
wherein X is phenyl or methyl and n is an integer from 1 to 15.

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51. A method of claim 49 or 50 wherein the effective amount of the compound is from about  $2\mu\text{M}$  to about  $6000\mu\text{M}$ .

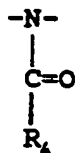
5 52. A method of claim 51, wherein the contacting is effected continuously for at least 48 hours.

53. A method of treating a patient having a tumor characterized by proliferation of neoplastic cells which comprises administering to the patient an amount of a compound effective to selectively induce terminal differentiation of such neoplastic cells and thereby inhibit their proliferation, the compound having the structure:



wherein each of A, A<sub>1</sub>, A<sub>2</sub>, A<sub>3</sub>, and A<sub>4</sub> represent a polar group which comprises a nitrogen, sulfur or oxygen atom and wherein each of A, A<sub>1</sub>, A<sub>2</sub>, A<sub>3</sub>, and A<sub>4</sub> may independently be the same as, or different from, the others of A, A<sub>1</sub>, A<sub>2</sub>, A<sub>3</sub>, and A<sub>4</sub>;

wherein each of R and R<sub>1</sub> is a hydrogen atom; a lower alkyl, alkenyl, or alkenyl group; or a group having the structure:



wherein each of R<sub>2</sub> and R<sub>3</sub> being a hydrogen atom or a lower alkyl, alkenyl, or alkynyl group; and wherein each of R, R<sub>1</sub>, R<sub>2</sub> and R<sub>3</sub>;

wherein each of [R-A] and [A<sub>4</sub>-R<sub>1</sub>] have a dipolar moment greater than about 2.7 Debye units;

5 wherein each of B, B<sub>1</sub>, B<sub>2</sub>, and B<sub>3</sub> represents a nonpolar group of which comprises least 4 atoms in a chain, the termini of which chains are attached to A and A<sub>1</sub>, A<sub>1</sub> and A<sub>2</sub>, A<sub>2</sub> and A<sub>3</sub>, and A<sub>3</sub> and A<sub>4</sub>, respectively; wherein each such atom is oxygen, nitrogen, carbon, or sulfur  
10 and wherein each of B, B<sub>1</sub>, B<sub>2</sub>, and B<sub>3</sub> may independently be the same as, or different from the others of B, B<sub>1</sub>, B<sub>2</sub> and B<sub>3</sub>;

15 and wherein each of a and b is independently 0 or 1.

20

25

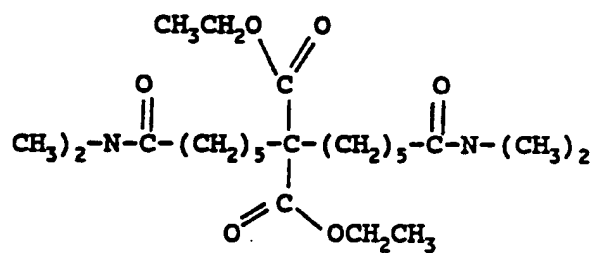
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98

54. A method of claim 53, wherein the compound has the structure:

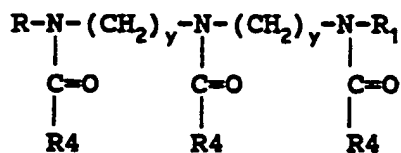
5



10

55. A method of claim 53, wherein the compound has the structure:

15



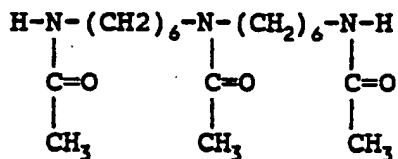
20

wherein each of R, R<sub>1</sub>, and R<sub>4</sub> is a hydrogen atom or a methyl, ethyl or propyl group, R and R<sub>1</sub> are the same and R and R<sub>4</sub> are the same or different; and wherein y is 4, 5, 6 or 7.

25

56. A method of claim 55, wherein the compound has the structure:

30

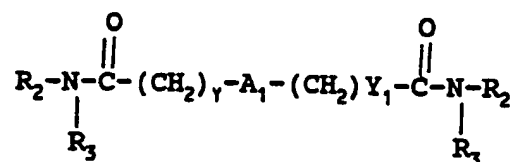


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57. A method of claim 51, wherein the compound has the structure:

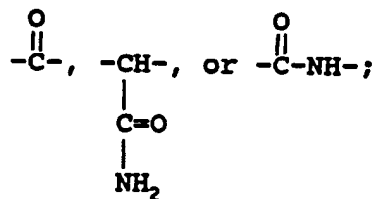
5



10

wherein each of  $\text{R}_2$  and  $\text{R}_3$  is a hydrogen atom or a methyl, ethyl or propyl group and each being the same or different; wherein  $\text{A}_1$  is

15

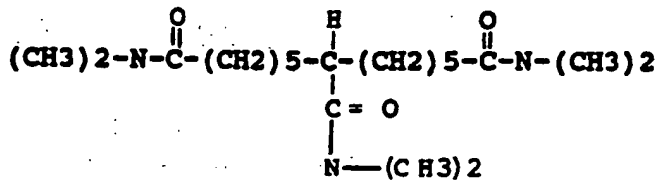


20

and wherein each of  $x$  and  $y$  is independently 5, 6 or 7.

58. A method of claim 57, wherein the compound has the structure:

25



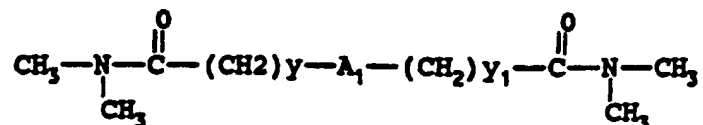
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59. A method of claim 57, wherein the compound has a structure:

5

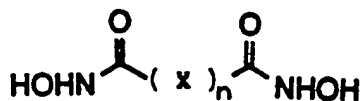


10

60. A method of treating patient having a tumor characterized by proliferation of neoplastic cells which comprises administering to the patient an amount of a compound effective to selectively induce terminal differentiation, which neoplastic cells and thereby inhibit their proliferation, the compound having the structure:

15

20



25

wherein X is phenyl or methyl and n is an integer from 1 to 15.

30

61. A method of treating patient having a tumor characterized by proliferation of neoplastic cells which comprises administering to the patient an amount of a compound of claims 23-39, in an amount effective to selectively induce terminal differentiation of neoplastic cells and thereby inhibit their proliferation.

35

62. A method of claim 60 or 61, wherein the amount of the compound is less than an amount which causes toxicity in the patient.

5

63. A method of claim 62, wherein the tumor is lung cancer, acute lymphoid myeloma, bladder melanoma, renal carcinoma, breast carcinoma or colorectal carcinoma.

10 64. A method of claim 60 or 61, wherein the administering is effected intravenously.

65. A method of claim 60 or 61, wherein the administering is effected orally.

15

66. A pharmaceutical composition comprising a pharmaceutically acceptable carrier and the compound of claim 1 or 23-39, in an amount effective to selectively induce terminal differentiation of suitable neoplastic cells and less than an amount which causes toxicity in a patient.

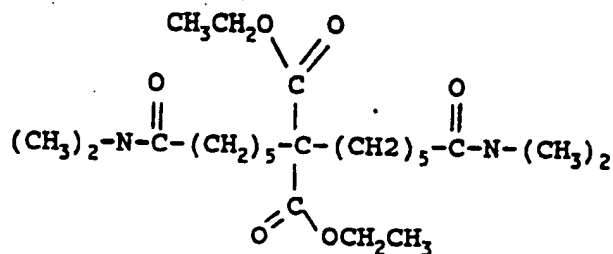
20

67. A pharmaceutical composition of claim 66, wherein the amount of the compound is from about 0.1 gm/m<sup>2</sup>/day to about 30 gm/m<sup>2</sup>/day.

25

68. A pharmaceutical composition of claim 66, wherein the compound has the structure:

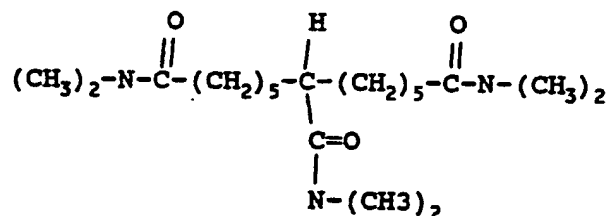
30



35

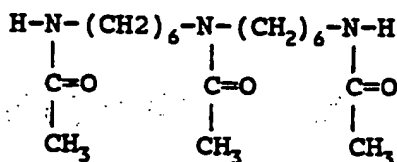
69. A pharmaceutical composition of claim 66, wherein the amount of the compound is not more than about 30 gm/m<sup>2</sup>/day.

70. A pharmaceutical composition of claim 66, wherein the compound has the structure:



71. A pharmaceutical composition of claim 70, wherein the amount of the compound is not more than about 30 gm/m<sup>2</sup>/day.

72. A pharmaceutical composition of claim 66, wherein the compound has the structure:

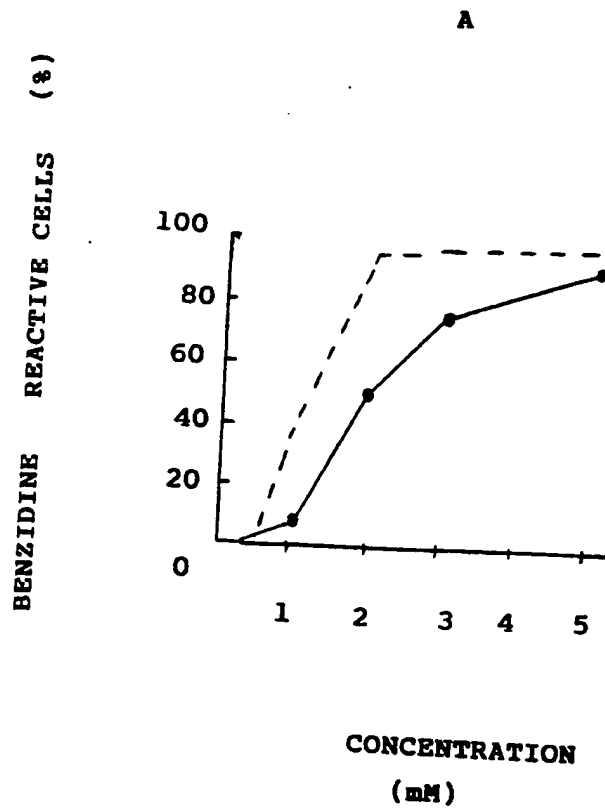


73. A pharmaceutical composition of claim 72, wherein the amount of the compound is not more than about 30 gm/m<sup>2</sup>/day.



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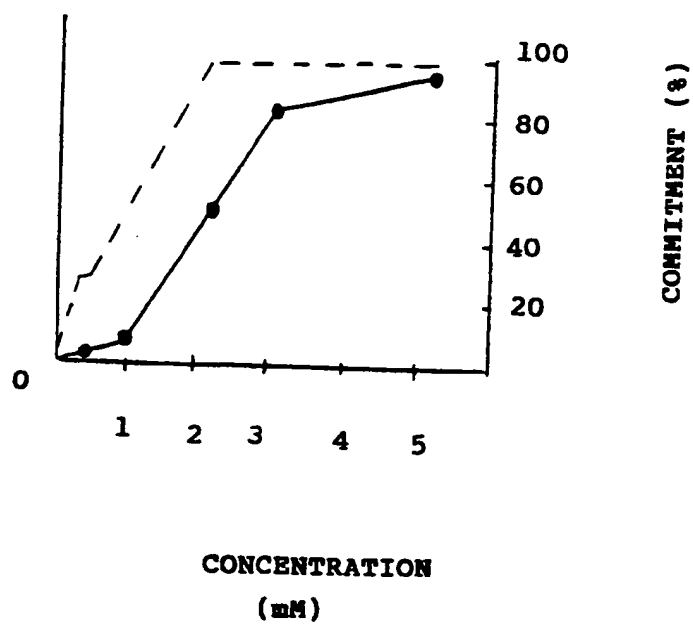
FIGURE 1A



2/12

FIGURE 1B

A

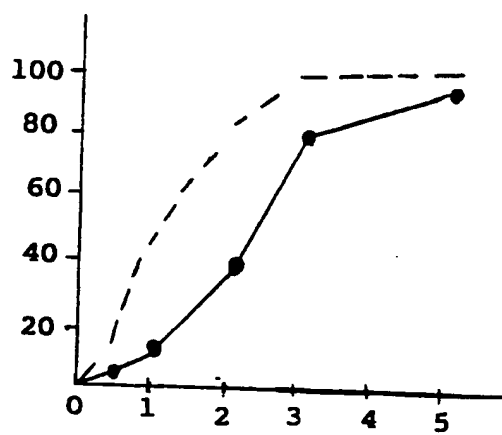


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FIGURE 1C

B

BENZIDINE REACTIVE CELLS (%)



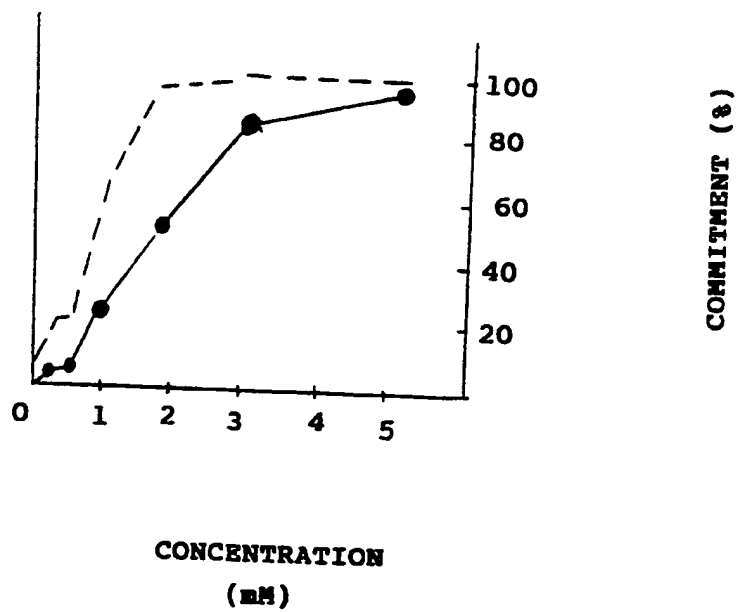
CONCENTRATION

(mM)

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FIGURE 1D

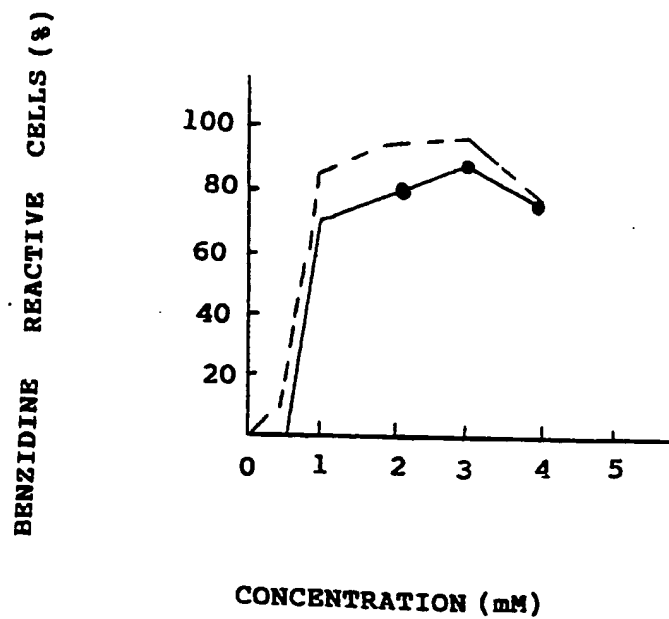
B



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FIGURE 1E

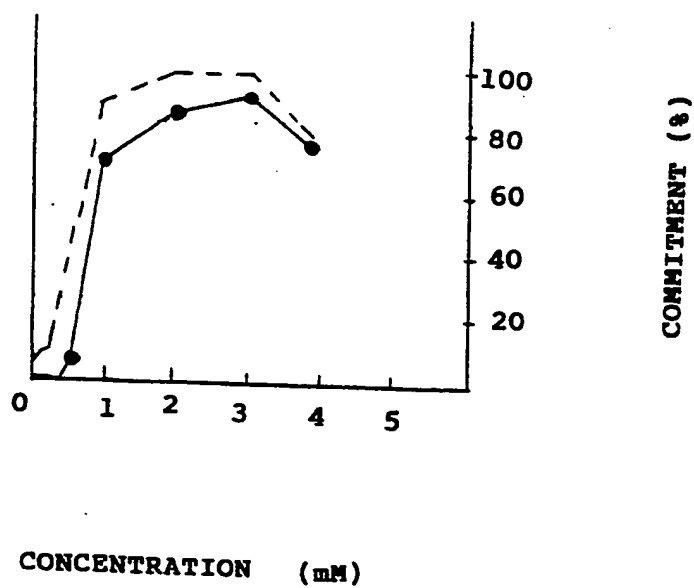
C



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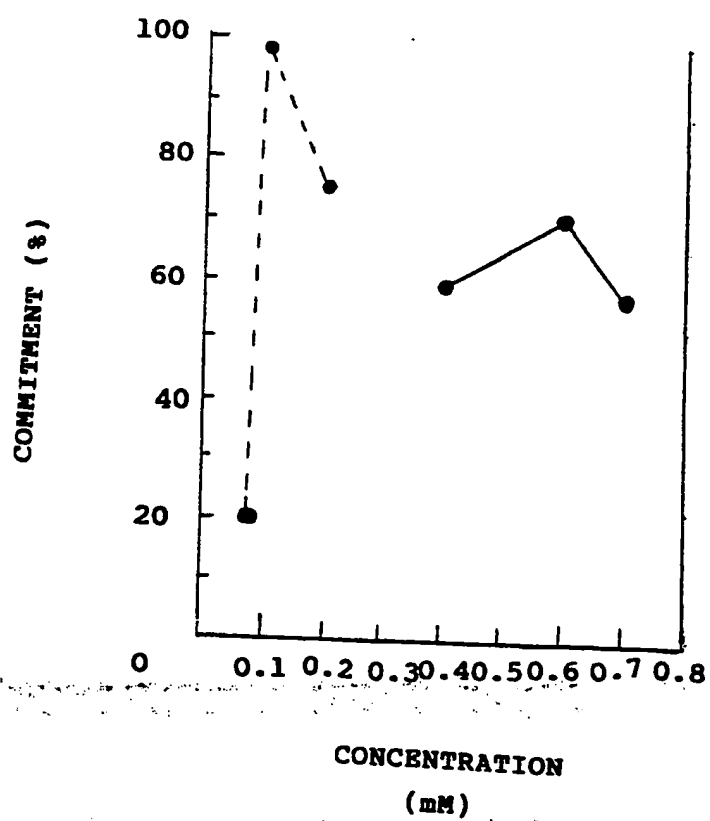
FIGURE 1F

C



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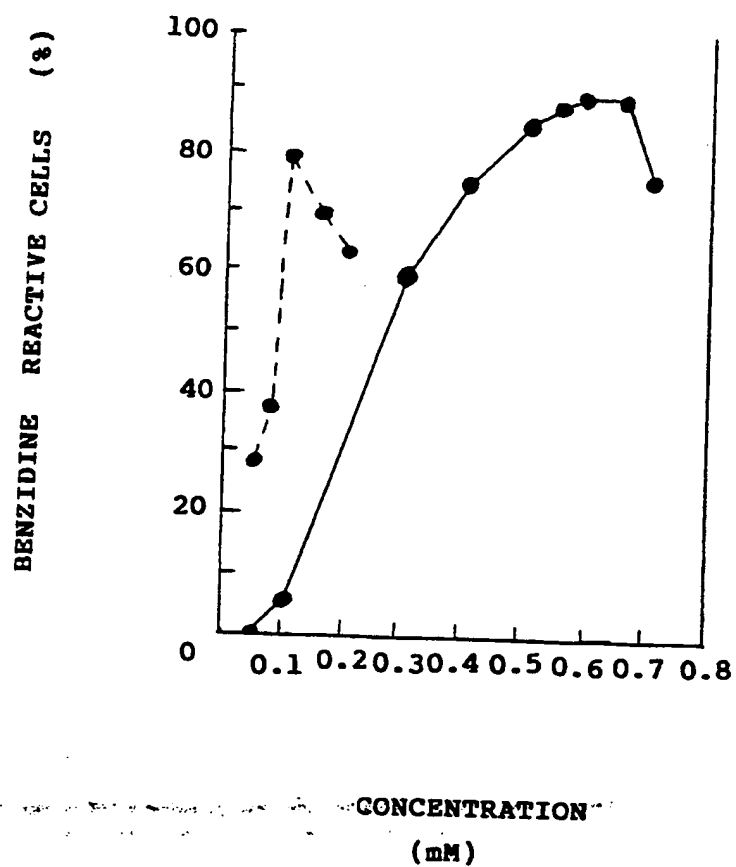
FIGURE 2A



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FIGURE 2B



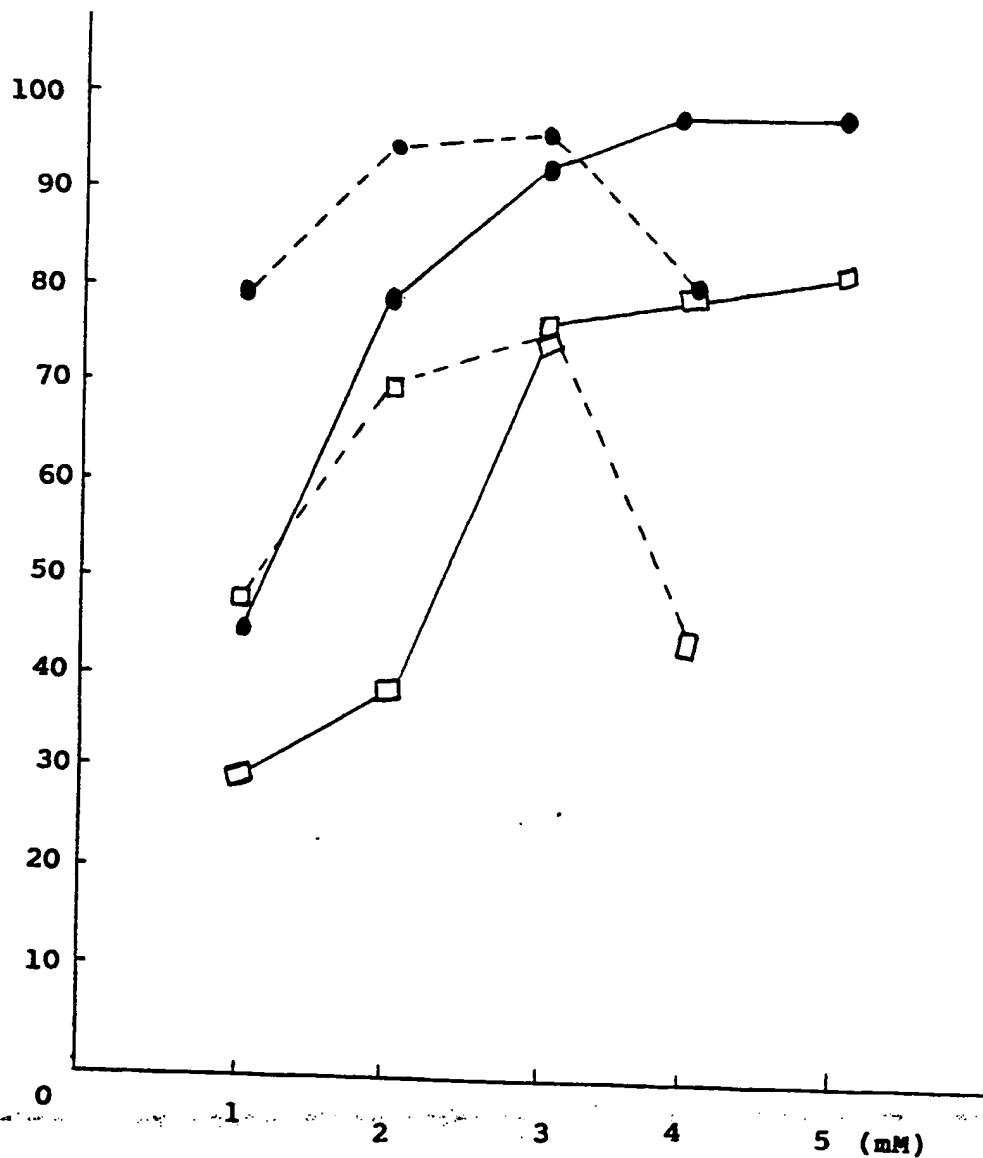
SUBSTITUTE SHEET



FIGURE 5A

COMMITMENT (%)

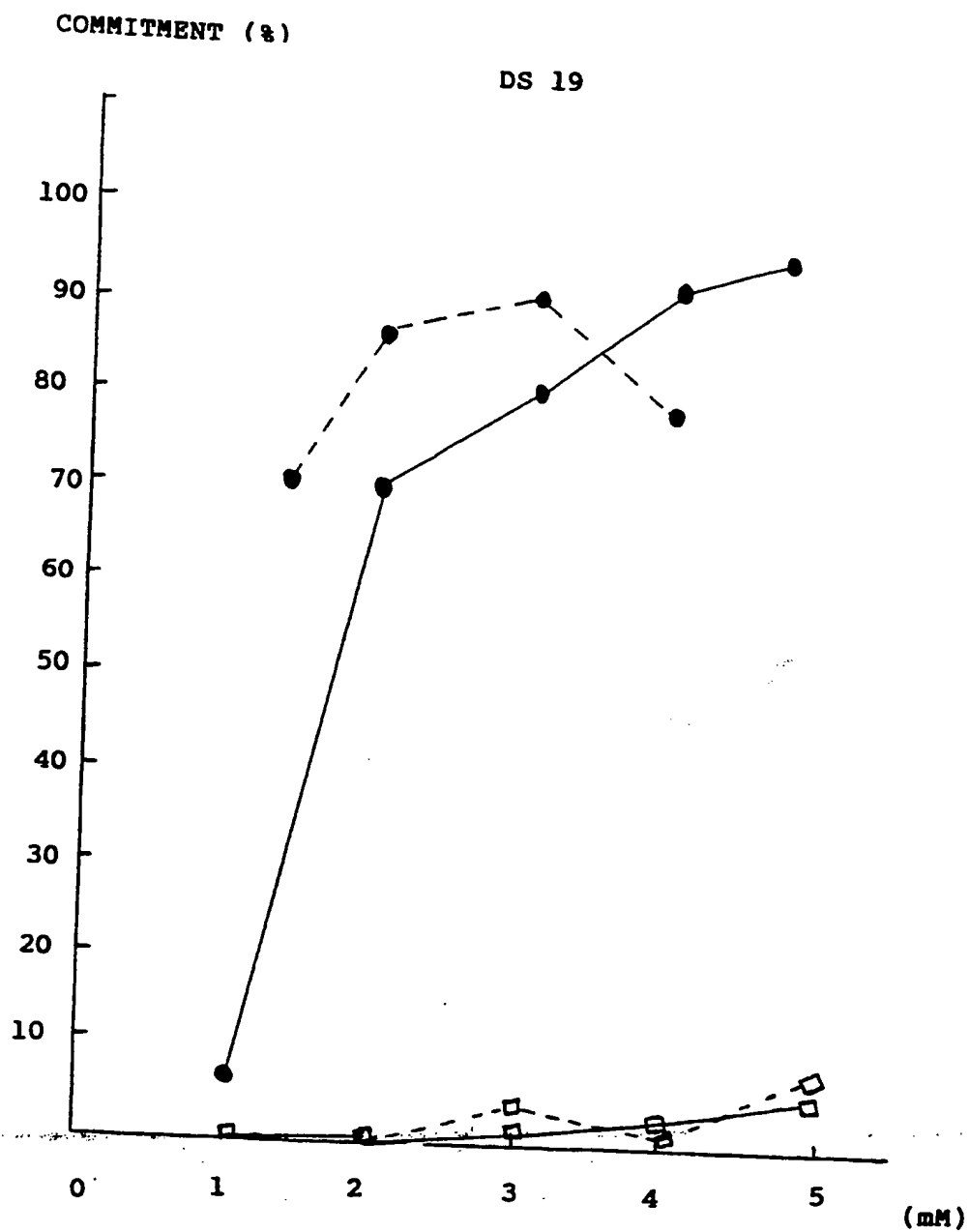
V 3-17



LEGEND: IC-135 @ 24 HRS.  $\square$  -  $\square$  HMBA @ 24 HRS.  $\square$  -  $\square$   
IC-135 @ 48 HRS.  $\bullet$  -  $\bullet$  HMBA @ 48 HRS.  $\bullet$  -  $\bullet$

SUBSTITUTE SHEET

FIGURE 5B



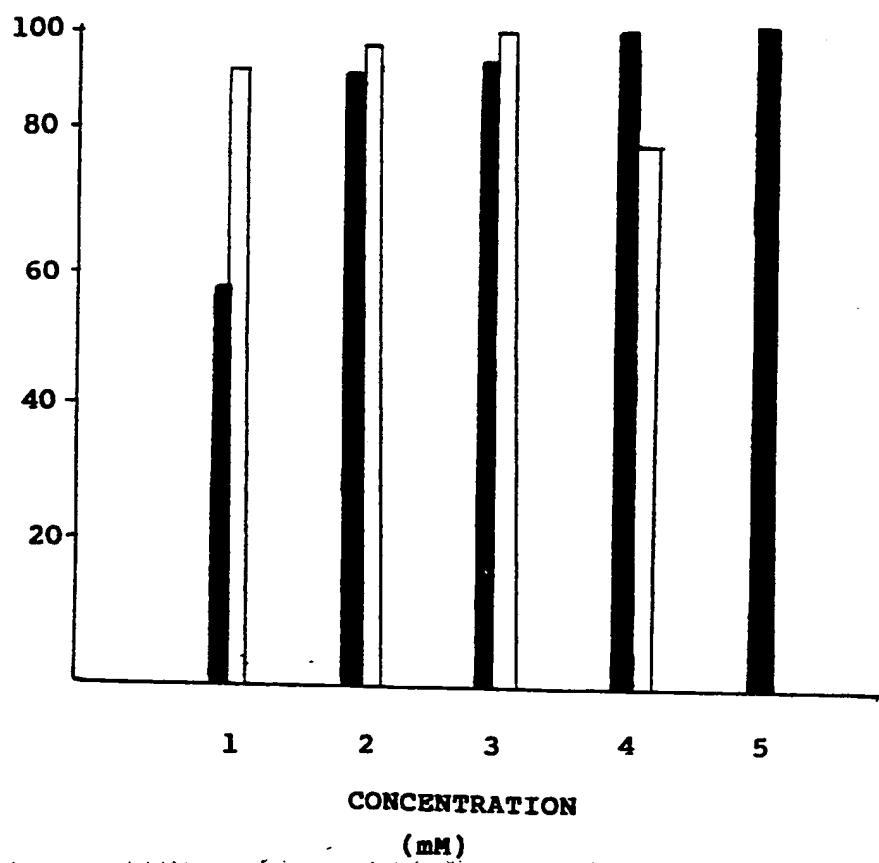
LEGEND: IC-135 @ 24 HRS. □-□ HMBA @ 24 HRS. □-□  
IC-135 @ 48 HRS. ●-● HMBA @ 48 HRS. ●-●

SUBSTITUTE SHEET

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FIGURE 6A

V3-17



LEGEND: HMBA



IC-135

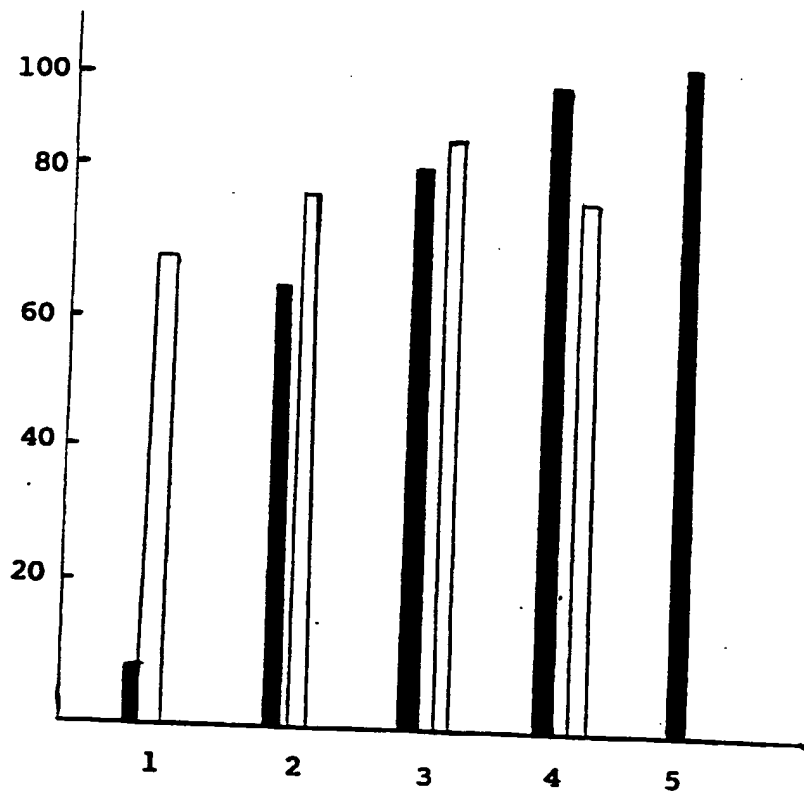


SUBSTITUTE SHEET

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FIGURE 6B

DS-19

CONCENTRATION  
(mM)

LEGEND:

HMBA

IC-135

SUBSTITUTE SHEET

# INTERNATIONAL SEARCH REPORT

International Application No

PCT/US90/02690

<b>I. CLASSIFICATION OF SUBJECT MATTER</b> (if several classification symbols apply, indicate all) <sup>3</sup>		
According to International Patent Classification (IPC) or to both National Classification and IPC		
IPC(5): 007C 55/02;69/34,76;233/04;235/20;243/14,18 007D 231/32;239/60;243/08;403/06		
US Cl/: 514/547,551,552;540/492;544/296,301;548/361;560/91,169,171;562/623;		
<b>II. FIELDS SEARCHED</b>		
Minimum Documentation Searched <sup>4</sup>		
Classification System	Classification Symbols	
US	514/547, 551, 552 540/492 544/296,301	548/361 564/126,148,149,153,160 560/91,169,171 193 562/623
Documentation Searched other than Minimum Documentation to the Extent that such Documents are Included in the Fields Searched <sup>5</sup>		
STN, CAS on line APS	U.S. Parent Applic	270,963 374,343 (Prior art)
<b>III. DOCUMENTS CONSIDERED TO BE RELEVANT</b> <sup>14</sup>		
Category <sup>6</sup>	Citation of Document, <sup>16</sup> with indication, where appropriate, of the relevant passages <sup>17</sup>	Relevant to Claim No. <sup>18</sup>
Y	US, A, 4,611,053 (SASA) 09 September 1986 see entire document	1,3-5,7,10-12, 28
P, Y	US, A, 4,801,748 (MURAHASHI et al) 31 January 1989 p5, No. 17	1,3-16
Y	US, A, 4,442,305 (WEITL et al) 10 April 1984 see entire document	1,3-16
Y	US, A, 2,346,665 (CURRY) 18 April 1944 p1 L 40-45	1,3-5,7,10-12,
X A	US, A, 2,279,973 (DIETRICH) 14 April 1942 p2 Col. 2 L 40-50	28 1,3-5,7,10-12
P,X	US, A, 4,863,967 (HALL et al) 05 September 1989 Col. 7 claim 1	1,30,66,67,69
<p><sup>19</sup> Special categories of cited documents:</p> <p>"A" document defining the general state of the art which is not considered to be of particular relevance</p> <p>"E" earlier document but published on or after the international filing date</p> <p>"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)</p> <p>"O" document referring to an oral disclosure, use, exhibition or other means</p> <p>"P" document published prior to the international filing date but later than the priority date claimed</p> <p>"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention</p> <p>"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step</p> <p>"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.</p> <p>"&amp;" document member of the same patent family</p>		
<b>IV. CERTIFICATION</b>		
Date of the Actual Completion of the International Search <sup>2</sup>		Date of Mailing of this International Search Report <sup>2</sup>
05 SEPTEMBER 1990		23 NOV 1990
International Searching Authority <sup>1</sup>		Signature of Authorized Officer <sup>20</sup>
ISA/US		BRIAN G. BEMBENICK

III. DOCUMENTS CONSIDERED TO BE RELEVANT (CONTINUED FROM THE SECOND SHEET)		
Category *	Citation of Document, <sup>14</sup> with indication, where appropriate, of the relevant passages <sup>17</sup>	Relevant to Claim No <sup>18</sup>
X	US, A, 3,450,673 (MCKILLIP) 17 June 1969 Col. 3 L 32-35	1,30
$\frac{Y}{X}$	Chemical Abstracts, Vol. 98, no. 23, "Antibypansomal and antimycotic effect of various hydroxamic acids". TABERNO et al. see abstract No. 191329v, Acta. Cient, Venez, 32(5) 411-16	<u>1,3-5,7,10</u> 33
$\frac{Y}{X}$	Arzneim-Forsch Vol 35, No 7, "Quantitative analysis of biological activities of diactinomycin analogs and methotrexate derivatives with Vander Waals volume" PRABHAKER et al, 1985 pages 1030-3	<u>1,3-5,7,10-12</u> 36 and 38
$\frac{Y,A}{X}$	Int. J. Cancer Vol 23, No 1, "Induction of Erythroid differentiation in murine erythroleukemia cells by nitrogen substituted polymethylene diamides", HOZUMI et al pp 119-22	1,3-5,7,10-12, 40,49,53,66 67,69 36 and 38
$\frac{Y}{X}$	Inorganic Chemistry, Vol 25, No 21, "Design of metal chelates with biological activity". BROWN et al, 1986, pp 3792-3796	<u>1,3-5,7,10-12</u> 28
$\frac{Y}{X}$	Chem Abstracts Vol 105, No 9, "A facile synthesis of aliphatic dihydroxamic acids of general formula RONR-CO-(CH ) -CO-NR'OR". BROWN et al, Synth. Comm. Vol 15, No 13, 1159-64, see abstract No 78501, 1985	<u>1,3-5,7,10-12</u> 28
$\frac{Y}{X}$	Chemical Abstracts Vol 101, No 7 "Synthesis of some dihydroxamic acid siderophores", DAS et al, J. Chem. Eng. Data, 29(3) 345-48, see abstract No 54665t 1984	<u>1,3-5,7,10-12</u> 28
A	"Organic Chemistry:", Morrison and Boyd, 3rd Ed. Allyn and Baron Inc. p 755	1,3-16
Y	Proc. Natl. Acad. Sci. USA, Vol 72 No 3 "Induction of Erythroid Differentiation ..." etc. TANAKA et al, March 1975, pp 1003-6	04-65
Y	J. Biol. Chem. Vol 253, No 12 "Inducers of Erythroleu- kemic Differentiation", REUBEN et al, June 1978 pp 4214-18	40-65
A	J. Org. Chem. Vol 46, No 25, "Lipophilic enterobac- tin analogs", WEITL et al, 1981, pp 5234-7	1,3-16

## FURTHER INFORMATION CONTINUED FROM THE SECOND SHEET

IPC (5) A61K 31/22, 23, 224  
 U.S. CL. 564/126, 148, 149, 153, 160, 193

V. ☐ OBSERVATIONS WHERE CERTAIN CLAIMS WERE FOUND UNSEARCHABLE<sup>1</sup>

This international search report has not been established in respect of certain claims under Article 17(2) (a) for the following reasons:

1. ☐ Claim numbers ..... because they relate to subject matter<sup>1</sup> not required to be searched by this Authority, namely:
  
2. ☐ Claim numbers ..... because they relate to parts of the international application that do not comply with the prescribed requirements to such an extent that no meaningful international search can be carried out<sup>1</sup>, specifically:
  
3. ☐ Claim numbers ..... because they are dependent claims not drafted in accordance with the second and third sentences of PCT Rule 6.4(a).

VI. ☐ OBSERVATIONS WHERE UNITY OF INVENTION IS LACKING<sup>2</sup>

This International Searching Authority found multiple inventions in this international application as follows:

1. ☐ As all required additional search fees were timely paid by the applicant, this international search report covers all searchable claims of the international application.
2. ☐ As only some of the required additional search fees were timely paid by the applicant, this international search report covers only those claims of the international application for which fees were paid, specifically claims:
  
3. ☐ No required additional search fees were timely paid by the applicant. Consequently, this international search report is restricted to the invention first mentioned in the claims; it is covered by claim numbers:
  
4. ☐ As all searchable claims could be searched without effort justifying an additional fee, the International Searching Authority did not invite payment of any additional fee.

## Remark on Protest

- ☐ The additional search fees were accompanied by applicant's protest.  
☐ No protest accompanied the payment of additional search fees.